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1       **A novel biotechnology based on periphytic biofilms with N-acyl-**  
2       **homoserine-lactones stimulation and lanthanum loading for**  
3       **phosphorus recovery**

4  
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1 **Abstract**

2 This study presents an approach for developing periphytic biofilm with N-acyl-  
3 homoserine-lactones (AHLs) stimulation and lanthanum (La, a rare earth element)  
4 loading, to achieve highly efficient and stable phosphorus (P) recovery from wastewater.  
5 AHLs stimulated biofilm growth and formation, also improved stable P entrapment by  
6 enhancing extracellular polymeric substance (EPS) production and optimizing P-  
7 entrapment bacterial communities. Periphytic biofilms loading La is based on ligand  
8 exchanges, and La loading achieved initial rapid P entrapment by surface adsorption.  
9 The combination of AHLs stimulation and La loading achieved 99.0% P entrapment.  
10 Interestingly, the enhanced EPS production stimulated by AHLs protected biofilms  
11 against La. Moreover, a method for P and La separately recovery from biofilms was  
12 developed, achieving 89-96% of P and 88-93% of La recovery. This study offers a  
13 promising biotechnology to reuse La from La-rich wastewater and recover P by biofilm  
14 doped with La, which results in a win-win situation for resource sustainability.

15 **Keywords:** Phosphorus recovery; Periphytic biofilm; N-acyl-homoserine-lactones;  
16 lanthanum loading

## 17 **1. Introduction**

18 The copious amounts of phosphorus (P) consumed in modern society (e.g., in foods and  
19 agricultural fertilizers) are eventually consolidated into wastewater (Cieřlik &  
20 Konieczka, 2017). Wastewater must be recognized as a valuable P resource from which  
21 P can be harvested to produce energy and raw materials (Guest et al., 2009). Recovery  
22 of P from wastewaters before discharge is essential to mitigate negative environmental  
23 impacts and achieve economic P sustainability (Conley et al., 2009; Wang et al., 2015).  
24 P recovery using microbial communities is an ambitious yet promising “green”  
25 approach to enhance P sustainability (Chen et al., 2019; Wu et al., 2018; Zeng et al.,  
26 2021).

27 Periphytic biofilms are ubiquitous microbial aggregates in waters and are capable of  
28 entrapping P (Wu, 2016). The P entrapment mechanism of periphytic biofilm have been  
29 explored systematically to improve the recovery process. Two main processes dominate  
30 P recovery: extracellular and intracellular entrapment. Extracellular P entrapment is  
31 favored by a high content of extracellular polymeric substances (EPS), as functional  
32 groups such as  $\text{NH}_4^+$  and mineral fractions such as Fe and Ca in the EPS are able to  
33 combine with P (Zhou et al., 2017). Intracellular P entrapment is carried out by a variety  
34 of P-entrapping microorganisms, especially those forming polyphosphate (Xu et al.,  
35 2020).

36 In periphytic biofilm, quorum sensing (QS), as a molecular communication system,  
37 relies on chemical signaling produced and transmitted by bacteria to direct and organize  
38 group behavior (Maddela et al., 2019). N-acyl homoserine lactones (AHLs), are the

39 main signal molecules used by the gram-negative bacteria prevailing in periphytic  
40 biofilms (Xu et al., 2020). AHLs-mediated QS has been shown to be closely related to  
41 EPS secretion, biofilm formation and P entrapment in periphytic biofilm (Lv et al., 2021;  
42 Xu et al., 2021). AHLs could regulate LysoPC metabolism that affects bacterial growth  
43 (Ma et al., 2018) and promotes EPS production by regulating the synthesis of amino  
44 acids and the UDP-Gal/UDP-Glc pathway (Tang et al., 2018b). In periphytic biofilm,  
45 AHLs addition up-regulated the genes involved in inorganic-P accumulation and P  
46 uptake (Xu et al., 2021). Thus, a strategy to promote P recovery through adjusting  
47 AHLs-mediated QS has been proposed.

48 Recently, lanthanum (La) has been used to develop P adsorbents, as it can enhance their  
49 rapid P adsorption capacity (Douglas et al., 2016). The 1-d P adsorption contents by La-  
50 modified bentonite were found to account for 86.4% of the 48-d adsorption in  $\text{KH}_2\text{PO}_4$   
51 solution (Haghseresht et al., 2009). The initial P entrapment rate of biofilms ( $0.5\text{-}1\text{ mg g}^{-1}\text{d}^{-1}$ )  
52 ( $\text{Lu et al., 2016}$ ) is poor compared to chemical adsorbents (e.g., drinking water  
53 treatment residue:  $2.42\text{ mg g}^{-1}\text{d}^{-1}$ ) (Wang et al., 2018). Therefore, La offers hope for  
54 improving the rapid P entrapment of biofilms. The periphytic biofilm could in principle  
55 be modified by loading La from La-rich wastewater (e.g., tailings wastewater, La  
56 extraction wastewater), which also attains the recovery of La (Gavrilescu, 2021).

57 The proposed P recovery technology, based on periphytic biofilms, aims to exploit the  
58 potential synergistic benefits of AHL stimulation and La loading. In this process, it is  
59 envisaged that enhanced production of EPS as promoted by AHLs will result in the  
60 protection of microorganisms to withstand direct La exposure (Liu et al., 2019; Sun et

61 [al., 2021](#)). The La loaded on the biofilm will provide additional binding sites for rapid  
62 sequestration of P ([Wang et al., 2018](#)), while AHLs stimulates biofilm growth and  
63 aggregation which, in turn, can enhance the La loading capacity of the biofilm. To  
64 develop this novel technology, an insight in the loading process of La on periphytic  
65 biofilm and the combined effects of AHLs and La on P entrapment is required.

66 The studies can be divided into four phases.

67 i. Periphytic biofilm was modified by AHLs stimulation and La loading and the  
68 effects of AHLs and La on the physiological properties of the biofilm were  
69 investigated.

70 ii. Following the periphytic biofilm development, two simulated P wastewater  
71 removal experiments based on inorganic P and organic P were performed to  
72 evaluate P recovery by periphytic biofilm, as well as the advantages and  
73 possible synergistic effects of AHLs and La on P recovery.

74 iii. The loading mechanism of La on biofilms and the effect of La loading contents  
75 on the P recovery of biofilms were evaluated.

76 iv. A method was developed for the separate recovery of P and La from periphytic  
77 biofilms.

## 78 **2. Materials and methods**

### 79 *2.1. Preparation of AHLs, La and periphytic biofilm*

80 As described by Shaw et al ([Shaw et al., 1997](#)), AHLs were extracted from the  
81 supernatant of periphytic biofilm system. The main AHL signaling molecules from  
82 periphytic biofilms were determined, namely *N*-octanoyl-DL-homoserine lactone (C8-

83 HSL), *N*-(3-oxooctanoyl)-L-homoserine lactone (3OC8-HSL), *N*-dodecanoyl-DL-  
84 homoserine lactone (C12-HSL) (see supplementary materials). The synthetic AHLs (>  
85 97%) were purchased from Sigma-Aldrich (Singapore): C8-HSL, 3OC8-HSL, C12-  
86 HSL.

87 Lanthanum (III) chloride (LaCl<sub>3</sub>) (> 99.99%) were purchased (Sigma-Aldrich,  
88 Singapore).

89 Periphytic biofilm was collected from Xuanwu Lake, Nanjing, China and inoculated  
90 into culture systems containing Woods Hole culture medium (Xu et al., 2021). After  
91 about two weeks, the periphytic biofilm had matured - as indicated by a deep green  
92 color. Periphytic biofilms were collected and washed two times with sterile NaCl  
93 solution (0.9 %) and centrifuged before tests were run.

#### 94 *2.2. Periphytic biofilm with AHLs stimulation and La loading*

95 An exogenous AHL mixture (C8-HSL, 3OC8-HSL, C12-HSL) was added to the  
96 periphytic biofilms system (biofilm concentration: 10 g /L wet weight), to yield a final  
97 AHLs concentration of 1 μM (for each AHL). Periphytic biofilms without added AHLs  
98 served as control. The systems were incubated under a standard light–dark cycle of 12  
99 h/12 h (xenon lamp, 150 W) at 25 ± 1°C in a dedicated incubation room for 7d to obtain  
100 AHLs-stimulated periphytic biofilm.

101 The La loading by AHLs-stimulated periphytic biofilm was based on LaCl<sub>3</sub> solution.  
102 AHLs-stimulated periphytic biofilm (10 g/L) in conical flasks were supplemented with  
103 La solution (2 mg La/L); raw periphytic biofilm (without added AHLs) served as the  
104 La group. All tests were run in triplicate. After 24h of La loading, the periphytic biofilm

105 from, the AHLs-La and La groups were collected. The samples were snap-frozen with  
106 liquid nitrogen and stored at  $-80\text{ }^{\circ}\text{C}$  for subsequent analyses of the main structures and  
107 the chemical compositions, EPS, biomass, enzymatic activities and community  
108 composition.

### 109 *2.3. Characteristics of periphytic biofilm*

110 After AHLs stimulation and La loading, periphytic biofilms were sampled for growth  
111 analysis based on the biomass. The main structures and the chemical compositions of  
112 periphytic biofilms were characterized by scanning electron microscopy (SEM, JEOL  
113 Co, Ltd., Japan) and energy dispersive X-ray spectroscopy (EDX, Oxford Instruments,  
114 UK). The contents of La, Fe, Al, Ca, K, and Mg in periphytic biofilm were quantified  
115 using inductively coupled plasma-atomic emission spectrometry (ICP-AES, Agilent  
116 Technologies, Santa Clara, CA).

117 Enzymatic activities of periphytic biofilm, including adenosine triphosphatase  
118 (ATPase), catalase (CAT), superoxide dismutase (SOD), acid phosphatase (ACP) and  
119 alkaline phosphatase (AKP) were determined with enzyme assay kits (WST-1 method,  
120 Jiancheng Bioengineering Institute, Nanjing, China). The diversity of the bacterial  
121 communities in the biofilm matrix were analyzed by MiSeq sequencing technology  
122 ([Morales Sergio & Holben William, 2009](#)). To study the P entrapment potential of the  
123 community, the changes in the relative abundances of P-entrapment bacteria in the  
124 different treatment groups were further analyzed.

### 125 *2.4. EPS analysis*

126 EPS were extracted from the periphytic biofilm with a sonication-cation exchange resin



127 method (Comte et al., 2006). The protein (PN) component of EPS was quantified by  
128 the coomassie brilliant blue staining method with bovine serum albumin as the standard  
129 (Frølund et al., 1995), whereas the polysaccharide (PS) component of EPS was  
130 determined using a phenol-sulfuric acid assay with glucose as the standard (Dubois et  
131 al., 1956).

132 The total contents of La, Fe, Al, Ca, K, and Mg in EPS were determined using ICP-  
133 AES. The chemical states of Al2p, Fe2p and C1s in EPS were determined by X-ray  
134 Photoelectron Spectroscopy (XPS, ESCALAB 250, Thermo Fisher Scientific, USA).

#### 135 *2.5. ROS accumulation of periphytic biofilm in the presence and absence of EPS*

136 In the presence and absence of EPS, the generation of reactive oxygen species (ROS)  
137 induced by La loading was measured. After AHLs stimulation, periphytic biofilm (0.5  
138 g wet weight) was put in conical flasks with La solution (100 mL, 5 mg La/L). To  
139 quantify ROS in the presence of EPS, biofilms were collected after 48h exposure. To  
140 quantify ROS in the absence of EPS, a heat treatment method (to avoid damaging  
141 biofilm cells) was used to removal EPS from periphytic biofilm prior to load La. In  
142 short, periphytic biofilm was collected, washed twice with 0.9% NaCl solution and  
143 centrifuged (10000 rpm, 10 min, 4 °C). Then the periphytic biofilm was resuspended  
144 with 0.9% NaCl solution and heated at 45 °C in a water bath for 30 min. All samples  
145 were centrifuged (10000 rpm, 10 min, 4 °C) to remove EPS from periphytic biofilm.

146 Previous studies have shown that limited heat treatment does not significantly affect  
147 ROS accumulation in periphytic biofilm (Fedyaeva et al., 2014; Li et al., 2018). The  
148 ROS were measured by reactive oxygen species Assay Kit method (100-500 T,  
149 Jiancheng Bioengineering Institute, Nanjing, China).

150 *2.6. P recovery test*

151 The P recovery tests included four groups, namely; (i) raw periphytic biofilm loaded  
152 with La (La group) and (ii) not loaded (Control group), (iii) AHLs-stimulated periphytic  
153 biofilm loaded with La (AHLs-La group) and (iv) not loaded with La (AHLs group).  
154 The P recovery test was based on  $\text{KH}_2\text{PO}_4$  and sodium glycerophosphate. Add 1 g (wet  
155 weight) of periphytic biofilm into a conical flask containing 100 mL of 5 mg P/L  
156 solution. In order to study the stability of P recovery by periphytic biofilms, the tests  
157 were run for 30 days to observe whether the P would be released from the biofilm after  
158 P recovery from the simulated wastewater. All tests were run in triplicate. pH in the  
159 periphytic biofilm systems was also measured (pH-10, Sartorius, Germany). The  
160 recovery tests were sampled regularly for P analysis. Total P analysis in solution was  
161 determined by ICP-AES.

162 In order to study the relationship between the La-load on periphytic biofilms and their  
163 P recovery capacity, the concentration of La in wastewater for the La loading test was  
164 set to 0.5 (AHLs-0.5 La), 2 (AHLs-2 La), 5 (AHLs-5 La) and 10 (AHLs-10 La) mg  
165 La/L. Samples of each were collected at the 24<sup>th</sup> h for La loading content analysis. The  
166 periphytic biofilms with different La loads were used for the P ( $\text{H}_2\text{PO}_4$ ) recovery test.  
167 P contents in periphytic biofilm were measured at 30d.

168 *2.7. P and La separately recovery processes*

169 A mixed solution of 3 %  $\text{H}_2\text{SO}_4$  and 3 %  $\text{HNO}_3$  were prepared. 1 g of periphytic biofilm  
170 was placed in a polypropylene centrifuge tube and 10 mL of acid solution was added.  
171 The mixtures were then sonicated (Ultrasonic cell disruptor; PS-60AL; Ldbsonic,

172 China) with 20% power, sonicated for 3 s at 10 s intervals, and repeated 30 times on  
173 ice. The biofilm was removed by centrifugation and La and P in the supernatant was  
174 separated by using a strong-acid cation exchange resin (Dowex 50X8; Dow Chemical  
175 Company, USA). A P-enriched solution and a La-enriched solution were obtained  
176 successively and P and La concentrations were measured using ICP-MS. The recovery  
177 efficiency was calculated based on the P entrapment amounts and REE loading amounts  
178 from ICP-MS data. The scheme of the recovery processes is illustrated in Fig. 1.

179

## 180 2.8. *Statistical analysis*

181 Statistically significant differences between the treatments and the controls were  
182 evaluated using ANOVA. For all analyses the significant *p*-value was set at 0.05.  
183 Pearson correlation between the EPS components and contents, and La loading capacity  
184 of EPS were evaluated using the package ‘vegan’ in R. The figures were drawn with  
185 Origin 9.0 and R software.

186

## 187 3. Results and discussion

### 188 3.1. *P recovery by periphytic biofilm with AHLs stimulation and La loading*

189 The P entrapment from simulated wastewater based on  $\text{KH}_2\text{PO}_4$ , is shown in Fig. 2a.  
190 La-loading substantially enhanced the initial entrapment by periphytic biofilms in  
191 comparison with non-La-loaded biofilms. According to the pseudo-second-order rate  
192 equation (see supplementary materials), from lowest to highest, the values were 0.63  
193  $\text{mg g}^{-1}\text{d}^{-1}$  (Control), 1.59  $\text{mg g}^{-1}\text{d}^{-1}$  (AHLs), 6.07  $\text{mg g}^{-1}\text{d}^{-1}$  (La) and 7.52  $\text{mg g}^{-1}\text{d}^{-1}$

194 (AHLs-La). Initial P entrapment rates were substantially increased by pre-loading with  
195 La, while AHLs stimulation had less effect on the initial P-entrapment rate. After the  
196 10th day, the P-entrapment content of periphytic biofilm in the AHLs-La group had  
197 reached 9.9 mg/g, i.e., 99% P-entrapment (Fig.2a). The P-entrapment of the La group  
198 had reached only 8.0 mg/g, i.e., 80% of the P-entrapment attained by the combination  
199 of La and AHLs. Moreover, the P-entrapment content by periphytic biofilm with AHLs-  
200 stimulation was found to steady at 9.9 mg/g over time until the 30th day, and the P  
201 concentration in wastewater had been below 0.1 mg/L. This indicated that the biofilm  
202 with AHLs-stimulation had a stable long-term P immobilization capacity, and it was  
203 difficult for P to be released into the wastewater again. Therefore, La-loading enhanced  
204 the rapid P-entrapment by periphytic biofilm, while AHLs stimulation tended to  
205 improve the stable P-entrapment.

206 For organic P wastewater, both AHLs and La promoted the entrapment of sodium  
207 glycerophosphate by the periphytic biofilms (Fig. 2b  $p < 0.01$ ). P entrapment was  
208 further indicted by the observation that AHLs and La enhanced acid/alkaline  
209 phosphatase activity of the biofilms (Fig.6).

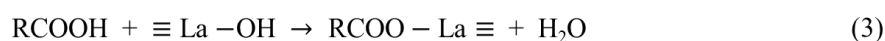
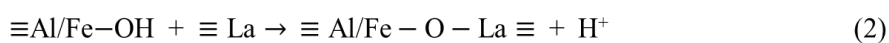
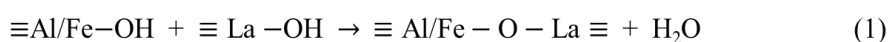
210 Whether it is inorganic P wastewater or difficult-to-treat organic P, the integration of  
211 AHLs-stimulation and La-loading has successfully improved the P recovery using  
212 periphytic biofilms, realizing the advantages of initial rapid P-entrapment, and steady  
213 and efficient P recovery. A method for P and La to be recovered separately from  
214 periphytic biofilms was developed; 88-92% of P and 87-91% of La can be recovered  
215 (Fig. 3), achieving sustainability of La and P.

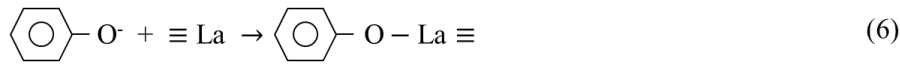
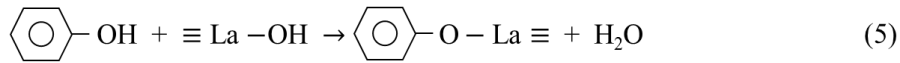
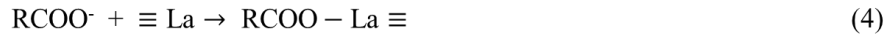
216 3.2. Relationship between La loading and P entrapment

217 3.2.1. La loading mechanism

218 A large proportion of La loaded on the EPS, which indicated that EPS may be the main  
219 place for La loading (Table 1). The strong positive correlation between EPS production  
220 and La loading content supported this hypothesis (see supplementary materials). The  
221 chemical species of Al2p, Fe2p and C1s on EPS with and without La loading are shown  
222 in Table 1.

223 Al/Fe oxides increased with increasing La-loading, while a decrease in Al and Fe  
224 hydroxides were clearly observed. C in periphytic biofilm included aromatic C/C-C/C-  
225 H, C-O/aromatic C, C=O/ketone C, C in carboxylate groups, and C in carbonate groups.  
226 As La-loading increased, aromatic C/C-C/C-H (284.5–284.9 eV) and carboxylate  
227 groups (288.3 eV) tended to decrease, and C-O/aromatic C (286–286.3 eV) groups  
228 tended to increase. These results suggested the occurrence of ligand exchanges of La  
229 with the Al/Fe hydroxides and the carboxylate/phenol groups. La speciation is governed  
230 by the La ion hydrolysis. As pH was not regulated in this study and varied between  
231 7.31–7.78, the dominant La species would be mostly La<sup>3+</sup> with a small proportion of  
232 LaOH<sup>2+</sup>; the existence of La(OH)<sub>2</sub><sup>+</sup>, La(OH)<sub>3</sub> and La(OH)<sub>4</sub><sup>-</sup> is unlikely in this pH range  
233 (Bouyer et al., 2006). Thus, La loading into EPS was mainly based on ligand exchanges  
234 (equations (1)–(6)).





### 241 3.2.2. *La loading enhances rapid P entrapment*

242 The effect of pre-loading with La on P entrapment of periphytic biofilms is shown in  
 243 Fig. 2c and 2d. Initial P-entrapment clearly increased as the La load increased for  
 244 groups AHLs-0.5 La to AHLs-10 La ( $p < 0.01$ ). In particular, the P-entrapment contents  
 245 rapidly increased to  $9.9 \text{ mg g}^{-1}$  in the AHLs-10 La group in which the initial P-  
 246 entrapment rate was up to  $32.5 \text{ mg g}^{-1}\text{d}^{-1}$  (see supplementary materials). The results  
 247 indicated that enhanced initial P-entrapment occurs as the La load increases in  
 248 periphytic biofilm.

249 La have remarkably high binding affinities for P. La are able to combine with P, forming  
 250  $\text{LaPO}_4 \cdot n\text{H}_2\text{O}$ , with a theoretical binding ratio between La and phosphate of 1:1 (Zhi  
 251 [et al., 2020](#)). In the simulated P wastewater, the dominant phosphate species at a pH  
 252 7.31-7.78 would be  $\text{H}_2\text{PO}_4^-$  and  $\text{HPO}_4^-$ , as these are the only two species present at a  
 253 pH value of 5-10 (Li [et al., 2020](#)). Phosphate and La combined rapidly on the biofilm  
 254 surface, and the complex species can be inferred to be as follows (Zhi [et al., 2020](#)):



### 257 3.3. *AHLs stimulation promotes biofilm growth and EPS production*

258 SEM indicated that a tight film and more substrate is present on the periphytic biofilm  
 259 surface after AHLs stimulation (see supplementary materials). The biofilm structure in

260 the control was observed to be looser and more porous. In the presence of AHLs for 7  
261 days, the biomass of periphytic biofilm increased by 37%, whereas in the control group,  
262 the biomass increased by only 8% (Fig. 4a). These results suggested that the addition  
263 of AHLs accelerated the aggregation and growth of the periphytic biofilm (Técher et  
264 al., 2020; Zhang et al., 2020).

265 AHLs also substantially stimulated EPS production (Fig. 4b and c). Specifically, PN  
266 and PS content of EPS increased by 76.0 mg/g, and 12.7 mg/g, respectively, compared  
267 with the control ( $p < 0.01$ ). This finding is in accordance with findings in previous  
268 studies in which supplementary QS signaling could increase EPS (Tang et al., 2018b).  
269 Previous studies have identified abundant functional groups (e.g., quaternary amines,  
270 tertiary amines) and mineral fractions (e.g.,  $Mg^{2+}$ ,  $Al^{3+}$ ,  $Ca^{2+}$ ,  $Fe^{3+}$ ) in EPS, all of which  
271 can complex with phosphate-P (Zhou et al., 2017). The increase of EPS production  
272 induced by AHLs promoted extracellular P entrapment.

273 Periphytic biofilm with AHLs-stimulation contained more Fe, Al, Ca, K, and Mg ( $p <$   
274  $0.01$ ) than the control (Table 1). This may be due to the increased biomass and EPS  
275 content upon AHLs stimulation, which provided more binding sites for metal. However,  
276 La loading may reduce the metal adsorption capacity due to a dilution effect. As  
277 expected, total metal contents decreased in periphytic biofilm after La loading,  
278 suggesting that La loading reduced the metal adsorption capacity by occupying the  
279 adsorption sites (equations (1)–(6)).

280

281 *3.4. Effect of AHLs and La on P-entrapment bacteria communities*

282 The microbial composition of the biofilms changed dramatically upon AHLs-  
283 stimulation (Fig. 5a). In contrast, La had a relatively small effect on the biofilm  
284 community. In agreement with previous studies, periphytic biofilms in all four groups  
285 principally consisted of the phyla *Proteobacteria*, *Bacteroidetes*, *Cyanobacteria*,  
286 *Firmicutes*, and *Planctomycetes* (Tang et al., 2019; Wang et al., 2020). AHLs  
287 stimulation increased the abundance of the *Proteobacteria* phylum (51.7-52.7%) such  
288 that it became the most dominant phylum under AHLs stimulation, while a decrease in  
289 *Cyanobacteria*. Such a shift in the microbial community structure may be attributable  
290 to the fact that some genera under *Proteobacteria* favor the AHLs in study as their QS  
291 signaling for communication.

292 Further in-depth analysis (Fig. 5b) showed the effect of AHLs and La on the P-  
293 entrapment bacterial communities. Nine reported P-entrapment bacteria were found in  
294 biofilms, such as *Acinetobacter* (Deinema et al., 1980), *Pseudomonas* (Tobin Karen et  
295 al., 2007) and *Aeromonas* (Lotter & Murphy, 1985). They entrapped large quantities of  
296 P and stored it as polyphosphate. With AHLs-stimulation and La-loading, the relative  
297 abundances of P-entrapment bacteria were 12.92% (AHLs-La group), 9.76% (AHLs  
298 group), 1.46% (La group) and 1.93% (Control group). AHLs-stimulation resulted in a  
299 big increase in the abundance of P-entrapment bacteria, which indicated that AHLs can  
300 be used as a beneficial signal molecule for P-entrapment bacterial communities to  
301 improve the competitiveness of P-entrapment bacteria (Scott et al., 2017; Youk & Lim,  
302 2014).

303 With no AHLs-stimulation, the La-loading led to a slight decrease in P-entrapment



304 bacteria abundance. While for AHLs-stimulated biofilms, La-loading had a significant  
305 positive effect on P-entrapment bacterial communities.

306

### 307 *3.5. AHLs stimulation alleviates the adverse bioeffects of La loading*

308 The biological effects of La in periphytic biofilm were evaluated by measuring ATPase,  
309 SOD and CAT activity (Fig. 6a). Microbial metabolic activity was represented by  
310 ATPase activity, which increased substantially after AHLs stimulation ( $p < 0.01$ ) (Liu  
311 et al., 2018). The SOD activity is an indicator of oxidative stress caused by adverse  
312 environmental conditions, and CAT alleviates damage caused by hydroxyl radicals (Xie  
313 et al., 2019). For the periphytic biofilm without AHLs stimulation, the SOD and CAT  
314 activity increased substantially with La loading ( $p < 0.01$ ). Interestingly, an increased  
315 SOD and CAT activity were not observed after AHLs stimulation, suggesting that AHLs  
316 stimulation mitigated the stress induced by La and protected microbial cells in  
317 periphytic biofilm.

318 In order to further explore the role of increased EPS productions caused by AHLs for  
319 the stress protection, EPS scavenging tests were conducted. La exposure induced  
320 adverse cell response: increased ROS levels. ROS prevalence substantially increased in  
321 the absence of EPS compared to ROS prevalence in EPS containing biofilms (Fig. 6b,  
322  $p < 0.01$ ), indicating that EPS can inhibit ROS prevalence. Moreover, in the presence  
323 of EPS, AHLs stimulation alleviated ROS prevalence caused by La loading. After EPS  
324 scavenging, no significant reduce of ROS prevalence was detected with AHLs  
325 stimulation ( $p > 0.05$ ). Therefore, enhanced production of EPS (Fig. 4b and c) caused

326 by AHLs-stimulation protected biofilms against La.

327 In the context of La loading, enhanced EPS production by AHLs stimulation has two  
328 benefits: to reduce or even eliminate direct La exposure on the periphytic biofilms thus  
329 providing sustainability for subsequent wastewater P recovery, and more importantly in  
330 this context, to provide a large number of binding sites that can load La (Li & Yu, 2014).  
331 EPS ineluctably influenced the distribution of La and their aggregation tendency which  
332 were important factors in their toxicity to microorganisms (Dang et al., 2018). With  
333 AHLs-stimulation, the elemental distribution investigation demonstrated that most La  
334 accumulated in the EPS (Table 1), which prevented La intrusion into the biofilm cells  
335 thus providing protection (Su et al., 2020). AHLs promoted EPS production and  
336 enhanced aggregation. The dense physical structure of the EPS acts as a barrier to La  
337 exposure in the interior of the periphytic biofilm. Aggregation also stabilizes microbial  
338 diversity, which enhance the ability of the community to adapt to La loading (Tang et  
339 al., 2018a).

340 Based on the results of the biological test and the P entrapment capacity of periphytic  
341 biofilm in aqueous solutions, AHLs stimulation and La loading on biofilms achieved P  
342 sustainability, and served to enhance the effectiveness of eutrophication control across  
343 a broad range of waters. Notably, La loading from La-rich wastewater also attained the  
344 La recovery.

#### 345 **4. Conclusions**

346 This study proposed an innovative bioaugmentation strategy based on periphytic  
347 biofilm to achieve highly efficient and stable P recovery from wastewater. AHLs-

348 stimulation enhanced stable P-entrapment via increasing the EPS production and  
349 optimizing P-entrapment bacterial communities. La-loading achieved initial rapid P  
350 entrapment by improving surface adsorption. Periphytic biofilm loads La from La-rich  
351 wastewater and realizes the reuse of La. Periphytic biofilm with AHLs stimulation and  
352 La loading achieved 99.0% P entrapment. Moreover, AHLs stimulation alleviated the  
353 adverse bioeffects of La loading. Finally, 88-92% of P and 87-91% of La can be  
354 recovered from periphytic biofilms, achieving sustainability of P and La.

355

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361

### 362 **Declaration of Competing Interest**

363 The authors declare that they have no known competing financial interests or personal  
364 relationships that could have appeared to influence the work reported in this paper.

365

### 366 **CRedit authorship contribution statement**

367 **Ying Xu:** Investigation, Methodology, Visualization, Data curation, Formal analysis,  
368 Writing – original draft. **Philip Kerr:** Formal analysis, Writing - review & editing. **Jan**  
369 **Dolfing:** Formal analysis, Writing - review & editing. **Bruce Rittmann:**  
370 Conceptualization, Writing - review & editing. **Yonghong Wu:** Funding acquisition,  
371 Supervision, Conceptualization, Writing - review & editing.

372

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## Figure captions

507

508 Figure 1. Dual process for recovery of P and La.

509

510 Figure 2. P entrapment performance of periphytic biofilms with AHLs-stimulation and  
511 La-loading. P entrapment by biofilms in wastewater based on (a)  $\text{KH}_2\text{PO}_4$  and (b)  
512 sodium glycerophosphate. (c) La loading by AHL-stimulated periphytic biofilm in  
513 solutions with different La concentrations. (d) P entrapment by AHL-stimulated  
514 periphytic biofilm with different La contents loaded. PB: periphytic biofilm; **AHLs-La**  
515 represents periphytic biofilm with AHLs-stimulation and La-loading; **AHLs** represents  
516 periphytic biofilm with AHLs-stimulation; **La** represents raw periphytic biofilm with  
517 La-loading; **Control** represents raw periphytic biofilm. Symbols indicate means and  
518 error bars show the standard deviation for the replicates (n=3).

519

520 Figure 3. Recovery of P and La from periphytic biofilm. Symbols indicate means and  
521 error bars show the standard deviation of the replicates (n=4).

522

523 Figure 4. Effects of AHLs-stimulation on (a) biomass of periphytic biofilm, (b) protein  
524 (PN) and (c) polysaccharide (PS) in EPS. Symbols indicate means and error bars show  
525 the standard deviation of the replicates (n=3). \*\* and \* indicate highly significant  
526 correlations ( $p < 0.01$ ) and significant correlations ( $p < 0.05$ ) respectively.

527

528 Figure 5. Effect of AHLs and La on microbial composition. (a) Community  
529 composition of bacteria at the phylum level, and (b) distribution of the P-entrapment  
530 bacteria in periphytic biofilm with AHLs-stimulation and La-loading.

531 Figure 6. Effect of AHLs and La on enzymatic activity and reactive oxygen species  
532 (ROS). (a) Effect of AHLs-stimulation and La-loading on enzymatic activity of  
533 periphytic biofilms. (b) In the presence or absence of EPS, the effect of La on the ROS  
534 accumulation in periphytic biofilms. Symbols indicate means and error bars show the

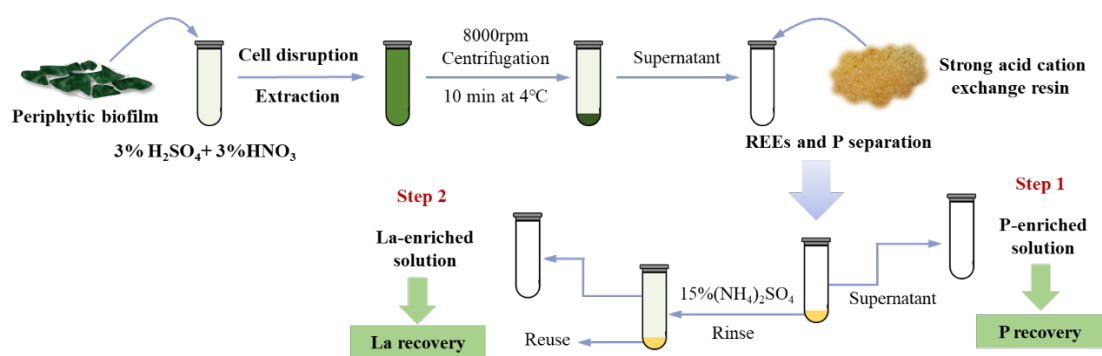


535 standard deviation of the replicates ( $n=3$ ). \*\* and \* indicate highly significant  
536 correlations ( $p < 0.01$ ) and significant correlations ( $p < 0.05$ ) respectively.

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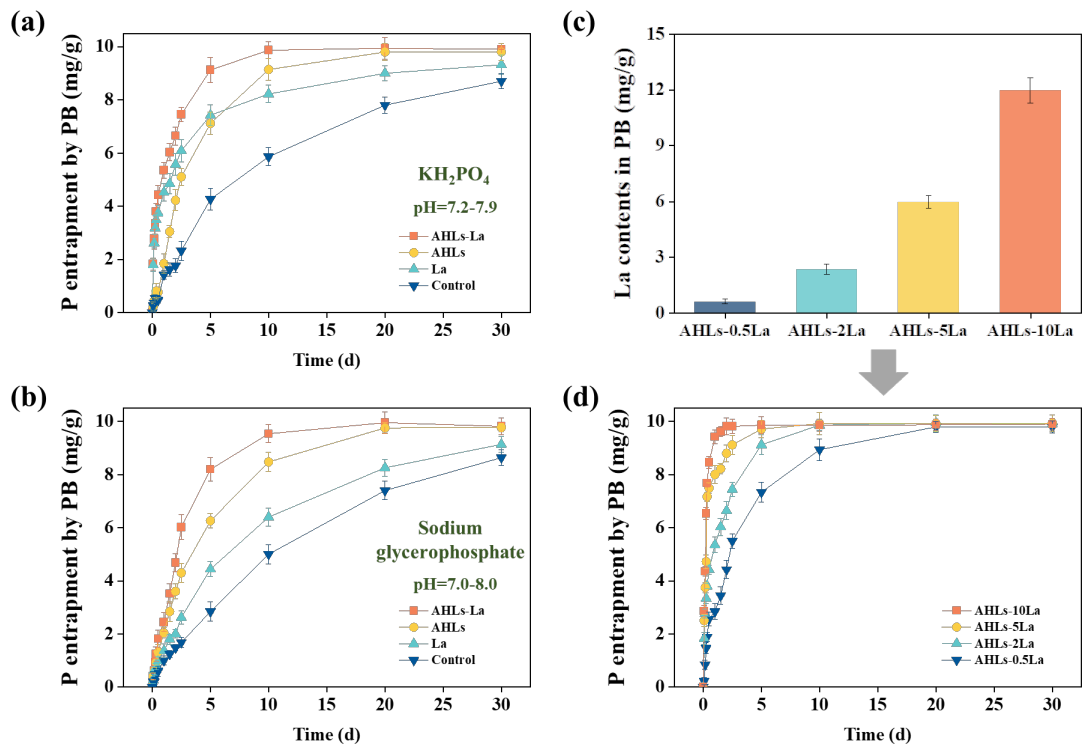
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Figure 1.

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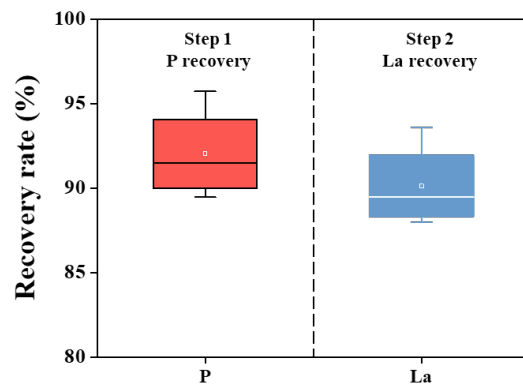
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Figure 2.

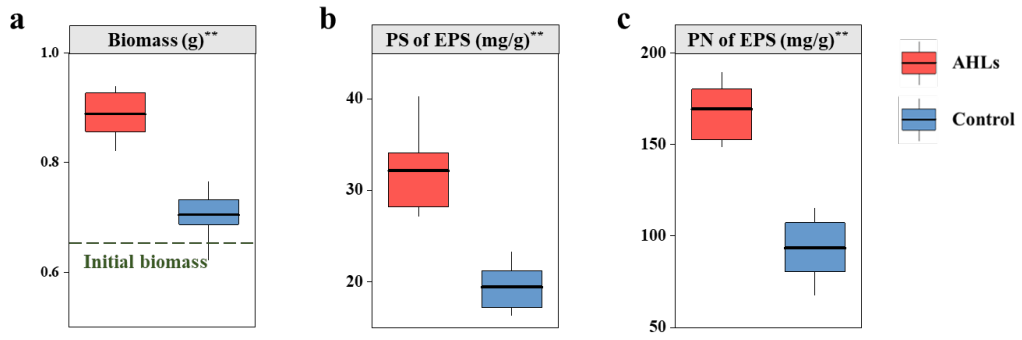


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Figure 3.



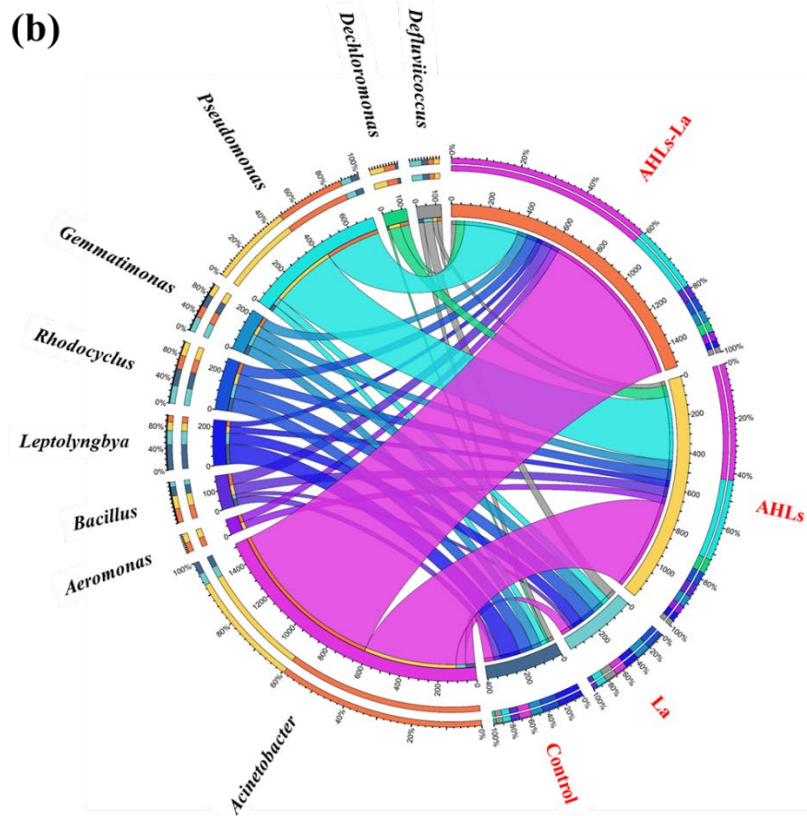
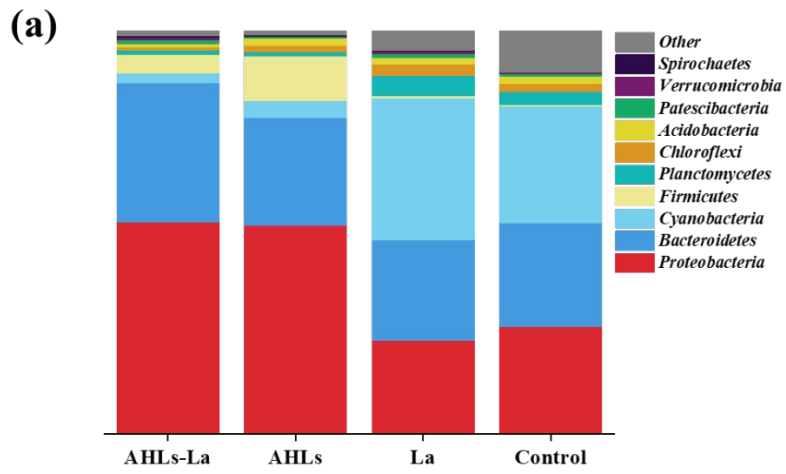
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Figure 4.



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Figure 5.

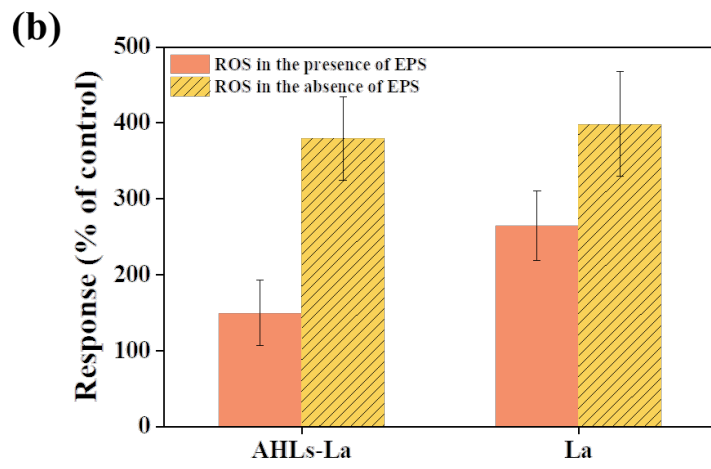
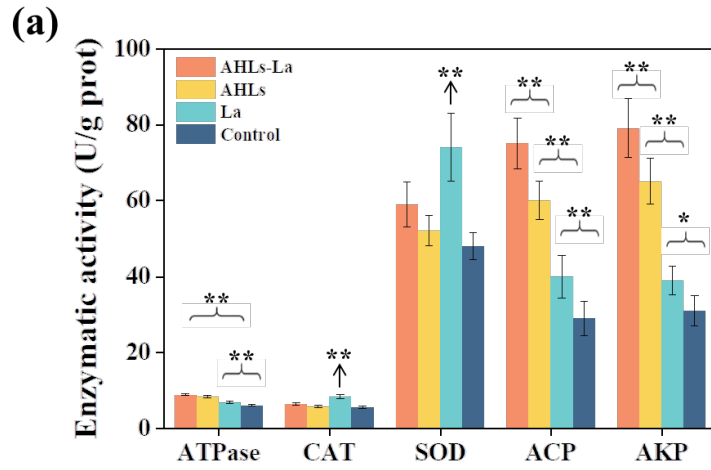


Figure 6.

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566 Table 1. The properties of periphytic biofilm and EPS with AHLs-stimulation and La-  
 567 loading.

Properties	AHLs-La	AHLs	La	Control	
pH	7.42±0.155	7.31±0.143	7.78±0.142	7.56±0.139	
Total contents in periphytic biofilm (mg/g) <sup>a</sup>	La	2.86±0.181	ND	1.74±0.136	ND
	K	7.42±0.635	8.19±0.530	7.03±0.492	7.27±0.346
	Mg	2.55±0.175	2.91±0.246	1.98±0.132	2.36±0.156
	Ca	18.18±0.620	18.43±0.518	8.40±0.359	8.99±0.461
	Fe	4.71±0.523	4.96±0.711	3.81±0.214	3.93±0.247
	Al	4.07±0.332	4.84±0.262	1.95±0.194	3.03±0.267
Total contents in EPS (mg/g) <sup>a</sup>	La	2.16±0.112	ND	1.04±0.096	ND
	K	3.49±0.127	4.12±0.173	3.09±0.190	3.22±0.201
	Mg	0.74±0.092	0.928±0.081	0.31±0.063	0.43±0.075
	Ca	7.03±0.512	8.09±0.701	4.26±0.223	4.77±0.345
	Fe	1.47±0.082	1.95±0.065	1.09±0.012	1.44±0.078
	Al	1.09±0.011	1.72±0.012	0.67±0.009	1.30±0.020
XPS (%)	Al <sub>2</sub> O <sub>3</sub>	58.33 (74.3) <sup>c</sup>	48.12 (74.3)	58.14 (74.4)	43.81 (74.4)
	Al(OH) <sub>3</sub>	41.67 (75.2)	51.88 (75.2)	41.86 (75.2)	56.19 (75.2)
	Fe <sub>2</sub> O <sub>3</sub>	71.60 (711.5)	40.31 (711.6)	60.92 (711.3)	47.28 (711.3)
	FeOOH	28.40 (713)	59.69 (713.6)	39.08 (713.6)	36.77 (713.4)
	Aromatic C/C-C/C-H	39.89 (284.5)	41.95 (284.9)	35.54 (284.8)	42.22 (284.7)
	C-O/aromatic C	44.51 (286)	36.96 (286.3)	45.62 (286.3)	36.36 (286.2)
C=O/ketone C	12.15 (287.5)	14.60 (287.5)	12.95 (287.6)	15.43 (287.5)	



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C in carboxy late group	3.44 (288.3)	6.49 (288.3)	5.89 (288.3)	5.98 (288.3)
C in carbona te group	0.001 (289.2)	0.001 (289.1)	0.001 (289.2)	0.001 (289.2)

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568 <sup>a</sup> n = 3; the detection limits were 0.005 mg/g for La, 0.005 mg/g for Fe, 0.020 mg/g for Al,

569 0.005 mg/g for Ca, 0.08 mg/g for K, and 0.002 mg/g for Mg, respectively.

570 <sup>b</sup> Not detectable.

571 <sup>c</sup> Data in parentheses show the binding energy (eV).

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