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A review of recent advances in adsorption desalination technologies

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Abstract

Adsorption-based desalination (AD) is an emerging concept to co-generate distilled fresh water and cooling applications. The present study is aimed to provide a comprehensive review of the adsorption desalination systems and subsequent hybridization with known conventional cycles such as the multiple-effect AD (MED), solar regenerable, integrated evaporator-condenser cascaded, and ejector integrated systems. The systems are investigated for energy consumption, productivity enhancement, and performance parameters, including production cost, daily water production, and performance coefficient. Comprehensive economic aspects, future challenges, and future progress of the technologies are discussed accordingly to pave researchers' paths for technological innovation. Traditional AD systems can produce specific daily water production of 25 kg per kg of adsorbent. The solar adsorption desalination-cooling (ADC) showed a promising specific cooling power of 112 W/kg along with a COP of 0.45. Furthermore, for a hybrid MEDAD cycle, the gain output ratio (GOR) and performance ratio (PR) is found to be 40%, along with an augmented water production rate from 60% to two folds. The AD technology could manage the

37 high salinity feed water with the production of low salinity water with a reasonable cost of
38 US\$0.2/m³.

39 **Keywords:** Adsorption desalination; adsorbent-adsorbate pairs; technologies; economic aspects;
40 system performance.

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62 **List of abbreviations**

63	\dot{Z}_k	capital investment cost of the component k
64	AD	adsorption desalination
65	ADC	adsorption desalination-cooling
66	ADEJ	adsorption cooling/ejector system
67	ADS	adsorption desalination system
68	ADVC	adsorption vapor compression
69	C	adsorption capacity (kg/kg)
70	C_e	adsorbate concentration at equilibrium (mg/L)
71	CFD	computational fluid dynamics
72	C_o	maximum adsorption capacity (kg/kg)
73	COP	coefficient of performance [-]
74	C_p	specific heat (kJ/kg)
75	CR	compression ratio
76	CRF	capital recovery factor
77	DEARC	double ejector refrigeration cycle
78	D_{so}	pre-exponential coefficient (m ² /s)
79	E_a	activation energy (kJ/mol)
80	EC	electrical conductivity (S/m)
81	EPA	environmental protection agency
82	ER	ejector refrigeration
83	EVCC	ejector vapor compression cycle
84	GOR	gain output ratio
85	g_{sw}	specific Gibbs energy of seawater
86	h	enthalpy (kJ/kg)
87	h_{fg}	average radius of adsorption particles (m)
88	HVCR	hybrid vapor compression refrigeration system
89	K	partition coefficient (L/g)
90	K_0	pre-exponential constant
91	KAUST	King Abdullah University of Science and Technology
92	LBT	low-brine temperature (°C)

93	LCC	life cycle costing (\$)
94	m	mass (kg)
95	M	molar mass (kg/mol)
96	\dot{m}	mass flow rate (kg/s)
97	MC	merocyanine
98	MED	multiple-effect distillation
99	MED-AD	multi effect desalination and adsorption desalination
100	MSF	multi-stage flash distillation
101	NF	nanofiltration
102	NUS	National University of Singapore
103	OCR	overall conversion ratio
104	ORC	organic Rankine cycle
105	PR	performance ratio
106	Q	heat energy (W)
107	q_e	adsorbed amount (mg/g)
108	RO	reverse osmosis
109	SCC	stress corrosion cracking
110	SCP	specific cooling power (TR/ton of adsorbent)
111	SDWP	specific daily water production (kg/kg adsorbent/day)
112	SP	spiropyran
113	T	temperature (K)
114	t	time (s)
115	t_{cycle}	cycle time (s)
116	TBT	top-brine temperature (°C)
117	TC	total carbon
118	TDS	total dissolved solids (mg/l)
119	TOC	total organic carbon
120	UPR	universal performance ratio
121	AD2EJ	adsorption desalination/2 ejectors system
122	UV	ultraviolet
123	VCC	vapor compression cycle

124	VCR	vapor compression refrigeration system
125	VCSC	vapor compression sub-cycle
126	ΔH_{ads}	isosteric enthalpy of adsorption
127	μ_0	chemical potential
128	τ	total annual time of system operation (hours)
129		
130	<i>Subscripts/superscripts</i>	
131	cu	copper
132	d	desalinated
133	des	desorption
134	i_{eff}	average annual effective discount rate
135	in	inlet
136	out	outlet
137	sg	silica-gel
138	sw	sea water
139		

140 **1. Introduction**

141 Water scarcity and its unbridled contamination are pressing issues that inscribe the world
142 population to curb this grave scenario. By 2050, around 6 billion population is expected to be
143 afflicted by water scarcity [1]. Freshwater is an exiguous resource that makes up only 0.75% of
144 the total quantity on earth readily available for human consumption, and 7.75% is stored in glaciers
145 and ice caps, while the remaining 97.5% is saline water [2]. The exponential population growth,
146 urbanization, rapid industrialization, climate change, and contamination of freshwater resources
147 pose a threat and lead towards a severe global challenge to ample provision of potable freshwater
148 [3]. It is high time to address the issue and consider the solemnness of posed challenges in this
149 situation. Existing resources of freshwater do not satisfy the water consumption patterns. It is
150 hence escalating the demand for freshwater. Alternative methods are employed to surmount the
151 challenges. It is rudimentary to address the water scarcity with saline water desalination.
152 Therefore, this method has been scientifically proved to be worth adopting, and the water industry
153 has become increasingly reliant on desalination techniques as it could be the most feasible option.
154 The desalination technique has evolved as a non-conventional alternate to counteract the challenge
155 of scarce freshwater resources by efficiently using abundant saline water resources [4]. The global
156 daily desalination capacity is 95 million cubic meters/day (MCM/d) or 38 billion cubic meters
157 (BCM/yr.) with an annual consumption of 75.2 TWh [5]. Researches have been carried out,
158 resulting in various membrane and thermal distillation systems to satisfy the water requirements.
159 According to a study by OECD [6], the global water demand is projected to escalate from 3500
160 km³ to 5500 km³ from 2000 to 2050, which records an increment of 55%, as presented in Fig. 1.
161 The manufacturing industry is found to be the primary cause of growing demand.

162 The variation in water utilization pattern disrupts the global water demand as a large amount
163 is returned to the mainstream after use. Thus, depending on the quality of water, the remaining
164 quantity is offered to be used downstream. It is also projected that by 2050, with the lack of new
165 policies, there will be a significant shift in water use, causing water demand [6]. The growing water
166 demand can be addressed by the desalination of seawater or treating the wastewater. Seawater
167 desalination can be the ultimate suitable process for some countries where direct wastewater reuse
168 is insufficient to fill the shortfall. In the past 20 years, the rise in the adoption of desalination
169 technology has opened doors to satisfy the global water demand. It is expected that the installation
170 and adoption of desalination technologies will be two folds by 2030, as shown in Fig. 2. Currently,

171 19,500 desalination plants are operating in 150 countries, having a production capacity of 100
 172 million m³/day, against the demand of 300 million people globally [7].

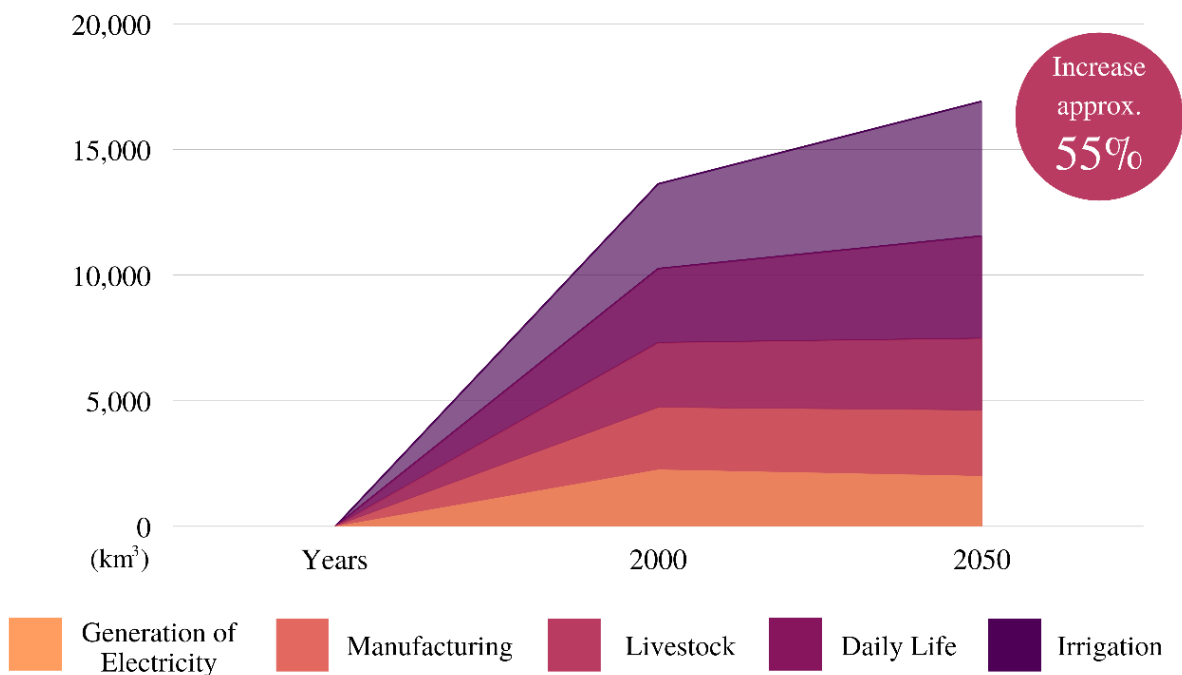


Fig. 1. Global blue water demand from 2000 to 2050, reproduced from [6].

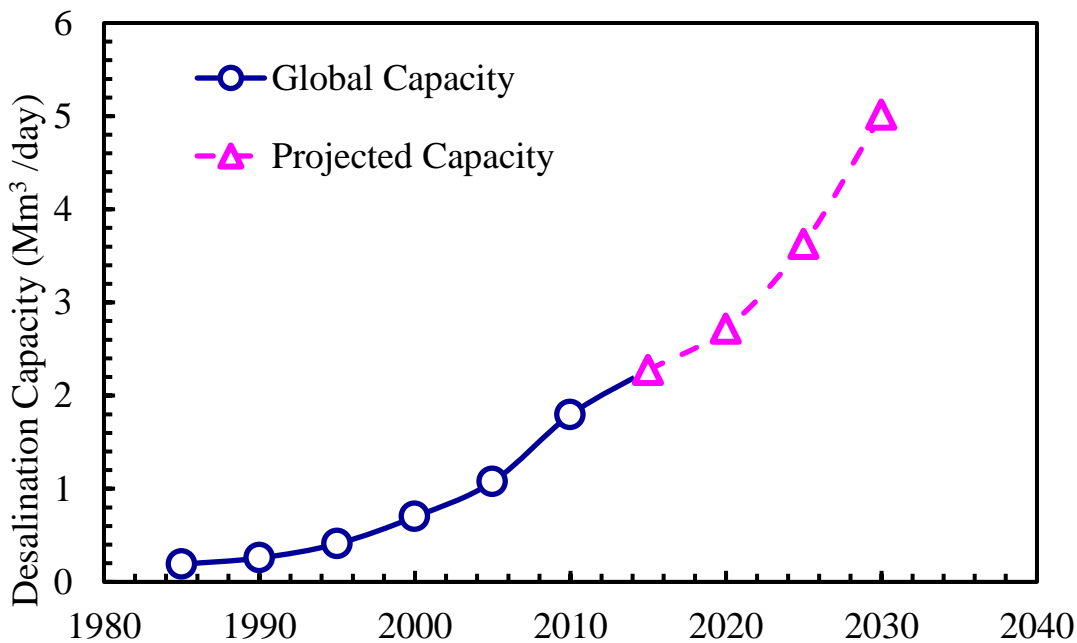
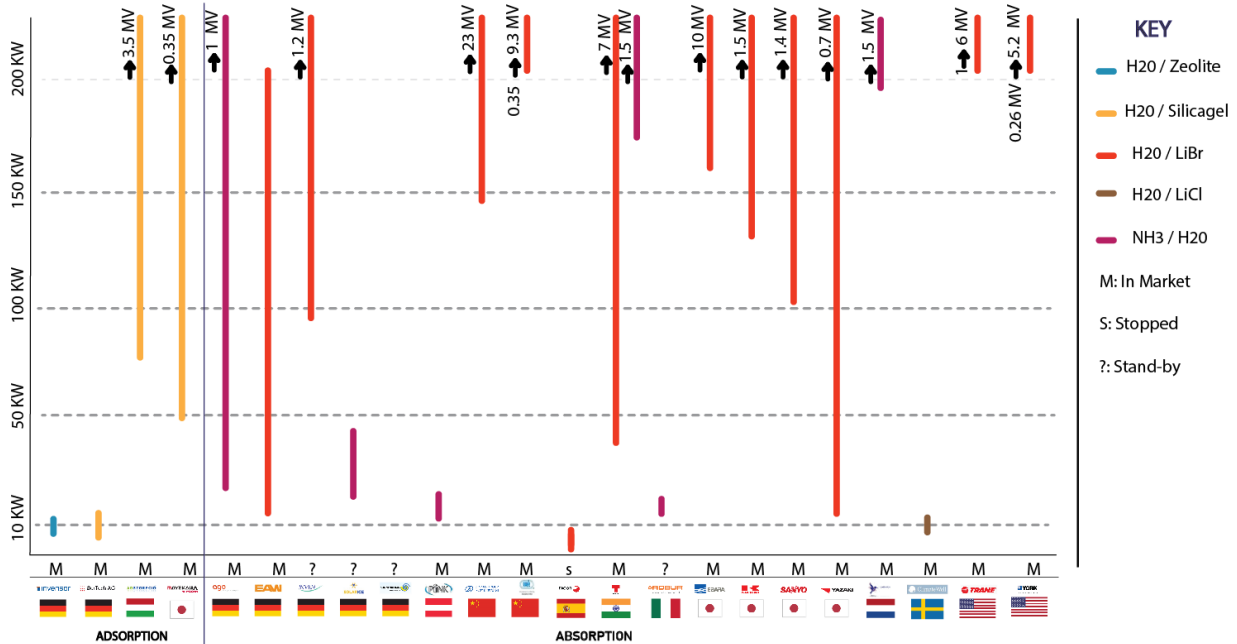


Fig. 2. Global desalination capacity of installed desalination technologies, trend, and projection from 1985 to 2030, reproduced from [7].

179 The technologies are classified, such as thermal desalination system that includes multi-stage
180 flash (MSF) or multi-effect distillation, providing around 18% and 17% of the global desalination
181 capacity. However, the membrane desalination systems include nano-filtration (NF) and reverse
182 osmosis (RO) that employ 3% and 69% of the global desalination capacity, respectively [8]. These
183 technologies are mainly adopted for large capacities [9], and most large-scale systems such as RO
184 and MSF are driven by fossil fuels having a large carbon footprint. The adsorption desalination
185 showed distinguished advantages over the other desalination options, which includes: (i) ability to
186 utilize solar heat or low-grade waste energy at a temperature lower than 100°C, (ii) reduced
187 maintenance cost due to simple construction and no significant moving parts, (iii) opportunity to
188 use commonly available environment-friendly adsorbent/adsorbate pairs, i.e., silica gel/water, (iv)
189 zero carbon footprint and emission reduction for greenhouse gases, (v) employment of controlled
190 corrosion and fouling rate on evaporator tubes material due to accruing of seawater evaporation at
191 comparatively reduced temperature (around < 35°C), (vi) the ability to cogenerate freshwater
192 along with cooling power, (vii) reduced electricity consumption of about 1.0-1.5 kW/m³ [10–15].
193 The refrigeration technologies, along with their manufacturers, are summarized in Fig. 3. Several
194 companies manufacture solar refrigeration machines using adsorption/absorption concepts with a
195 power rating ranging from 5 kW to 200 kW. The most commercialized systems possess the power
196 of less than 50 kW and employ H₂O/LiBr operating fluid and falling in the category of absorption
197 technology. The technologies mainly presented belong to German or Asian markets [16].
198



199

200 **Fig. 3.** A general representation of solar refrigeration technologies and associated power rating
 201 using adsorption/absorption concepts with their subsequent global manufacturers, reproduced
 202 from [16].

203

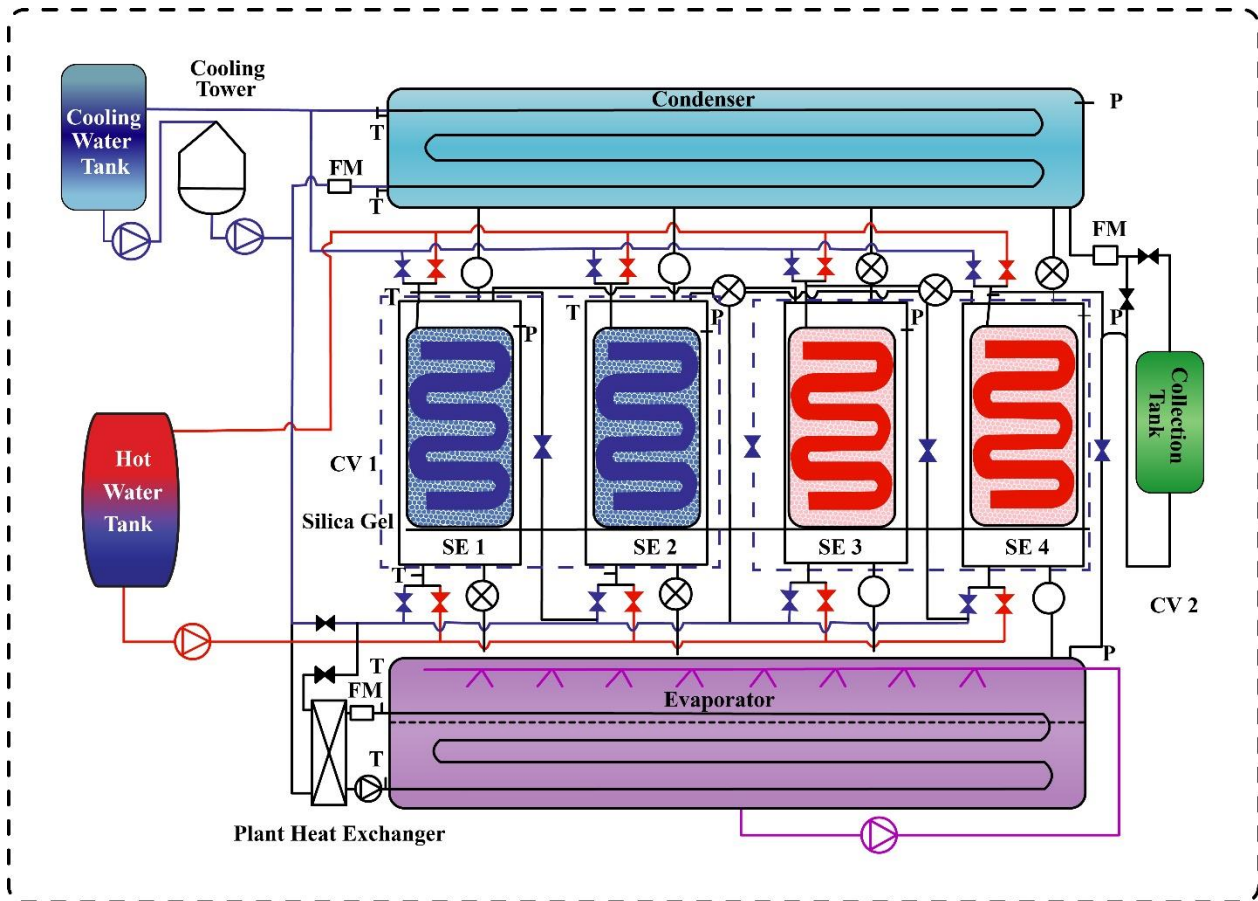
204 Apart from the benefits, the AD technologies possess some drawbacks of low system
 205 performance in terms of cooling capacity and COP. As a result, it has received little consideration
 206 from researchers in the past but has recently gained considerable importance because of its ability
 207 to be separable at low temperatures from driving heat source temperatures [17–20]. For example,
 208 desiccant air-conditioning (AC) involves adsorbate and water for the dehumidification process of
 209 process air [21,22]. The central systems included are desiccant [23–25], adsorption [26–28], and
 210 absorption technologies [29–32].

211 In 1984, one of the earliest units of adsorption desalination system was reported by Broughton
 212 [33][10]. It was a two-bed thermal adsorption desalination system for which simulations were
 213 performed. Several other developments were also made with the purpose of performance
 214 enhancement of the system. Zejli et al. [10,14] reported a multi-effect desalination (MED) system
 215 coupled with an adsorption heat pump utilizing zeolite/water pair. The heat pump coupled multi-
 216 effect desalination unit uses internal heat recovery to supply steam and seawater to the MED unit.
 217 A three-effect desalination system design with an evaporator sandwiched between two adsorption
 218 beds was proposed [10]. In this regard, the second bed utilizes the heat rejected from one bed

219 directly without processing because the second bed needs high-temperature thermal energy. Ng et
220 al. [10,13], and Wang and Ng [15] have presented the performance evaluation of a four-bed
221 adsorption desalination plant using silica gel/water. The same research group has reported specific
222 daily water production of 4.7 kg per kg of silica gel for an adsorption desalination plant utilizing
223 cooled water [13].

224 The adsorption desalination (AD) cycle works on two main processes: (i) the adsorption-
225 evaporation process and (ii) the desorption-condensation process [10,34–39]. In the first cycle of
226 the process, the adsorbent absorbs the vapors generated in the evaporator. Seawater is sprayed over
227 the tube bundle in the evaporator while the refrigerant circulation occurs [40]. The two significant
228 outputs are produced simultaneously by the adsorption cooling and desalination cycle. These
229 outputs are obtained using a multi-bed arrangement [10,41]. It is to note that the closed cycles of
230 air and water aids to significantly increase the energy recovery since the energy provided as input
231 is utilized to heat water and is preserved during the process [42]. The gas and water valve delay
232 ensures a conventional heat recovery scheme [43]. There is another type of energy recovery
233 scheme that involves both mass and heat recovery processes. The second category is attained
234 through the pressure equalization process for the pre-cooling and pre-heating beds. Hence, the
235 adsorption and desorption processes are both improved [15,43–46]. The brine lowers the input
236 energy requirement for feedwater heating, and then processed air is used for preheating water
237 within the dehumidifier. Heat recovery in an advanced AD cycle is attained either by the water run
238 around the condenser and evaporator or a device consisting of an enclosed evaporator and
239 condenser [47–50]. Despite the large consumption of energy, reduced productivity has gained
240 researchers' attention. Thereby, a few solutions are discussed accordingly in this study.

241

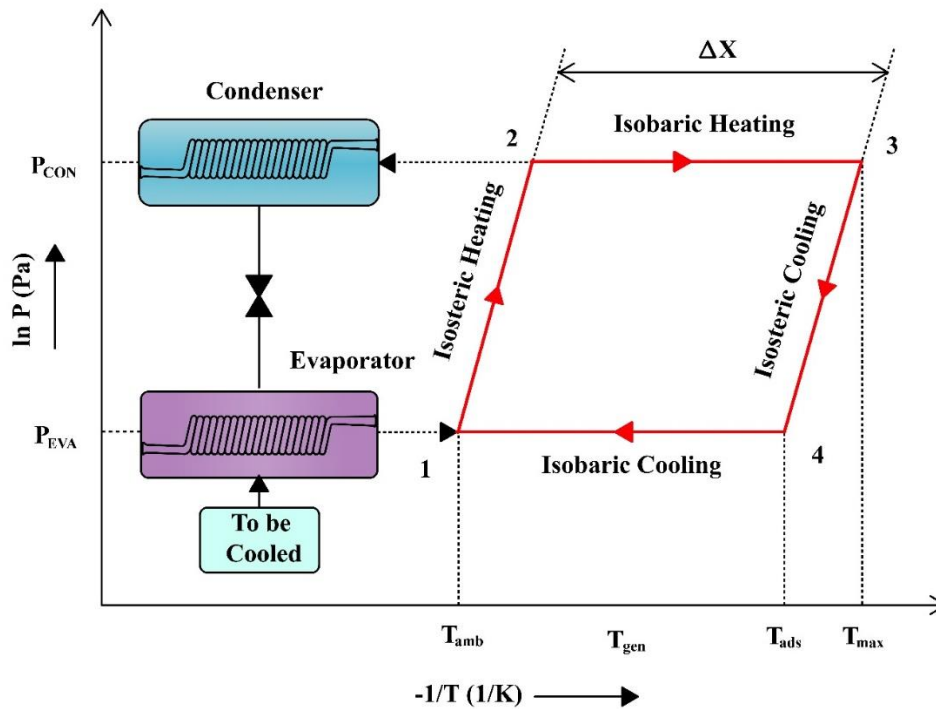


242

243 **Fig. 4.** Schematic diagram of a conventional adsorption desalination system.

244

245 The conventional adsorption desalination system is shown in Fig. 4, which consists of three
 246 major design components, i.e. (i) copper-nickel or stainless steel evaporator to prevent erosion, (ii)
 247 the reactor consisting of silica gel packed heat exchanger tubes, and (iii) condenser for water
 248 vapors condensation. The desalination system requires an intrinsic minimum available energy and
 249 is known to be an energy-intensive process [43,51–53].



250
 251 **Fig. 5.** A fundamental (PTX) diagram of an adsorption desalination thermodynamic cyclic
 252 process.

253
 254 A thermodynamic cyclic process for an adsorption desalination system is represented in Fig.
 255 5. First, the process from 1 to 2 undergoes an isosteric heating process, which goes to desorption
 256 from processes 2 to 3, also known as the isobaric heating process. Then, the isosteric heating occurs
 257 from 3 to 4, followed by adsorption, known as isobaric cooling. The PTX diagram is presented
 258 where $-1/T$ is plotted on the x-axis and $\ln P$ is plotted on the y-axis. Where X is the amount (kg of
 259 adsorbate per kg of adsorbent) of adsorbed adsorbate by the adsorbent at equilibrium conditions,
 260 it is helpful to depict the thermodynamic cycle of an adsorption desalination system (ADS), where
 261 X_{\min} , X_{\max} , T_{ads} , and T_{gen} represent the minimum and maximum amount of water adsorbed and the
 262 temperature at which the desorption and adsorption process takes place, respectively [54].

263 Adsorption working pairs, e.g., silica gel-water, are the necessitous components in the
 264 adsorption system. While comparing silica gel-water and zeolite-water adsorption working pairs,
 265 silica gel has a structure of dehydrated polymeric colloidal silica acid along with the pattern of
 266 $\text{SiO}_2 \cdot n\text{H}_2\text{O}$. The spherical particles of amorphous material are 2-20 nm in size, which form the
 267 silica by sticking with each other [55]. Silica gel is a preferred adsorbent due to its ability to take

268 up a reasonable amount of water (up to 40% by mass) [56] without considerable change in volume
269 or structure and capacity to release water when mildly heated [57]. Systems with two or more beds
270 are used in the literature [58] regarding the improvement in energy efficiency, and several studies
271 have also reviewed AD systems.

272 For instance, a technical review [58] is presented on MSF and MED desalination configurations
273 and processes, along with prevalent adsorption mechanisms and low-cost adsorbent materials. The
274 regeneration of saturated adsorbent and study of the physisorption or different mechanisms for
275 effective adsorption is recommended. Ng et al. [59] reported low-cost thermal desalination systems
276 based on pilot-scale experiments and life cycle cost analysis. The study recommends life cycle
277 costing (LCC) for the comparison of desalination technologies. Shahzad et al. [60] have also
278 studied the recent developments of the AD theory cycle and conventional MED-AD cycles. The
279 study analyzed different MED cycles while highlighting the critical role of AD cycle hybridization
280 with proven cycles, i.e., the MED cycle. For significant improvement in the yield of water
281 production, it is concluded that the UPR of the MED-AD cycle is the highest of all technologies
282 with furthermore projection of lowered water production LCC. Analysis of various AD systems
283 [10] for productivity improvements and proposed solutions for improving system and energy
284 efficiency is performed. Accordingly, recently developed working pairs used in different
285 configurations of AD systems are compared. Cooling power and water production mainly depend
286 on the adsorbent's adsorption rate and adsorptive ability, making the working pair a crucial design
287 parameter. Technological advancements in renewable energy sources assisted AD systems were
288 investigated [34]. It reports the different adsorption desalination-cooling systems, solar thermal
289 energy driven by ADS, waste heat-driven adsorption cooling cum desalination cycle, and AD cycle
290 with internal preheat recovery.

291 A review on adsorption working pairs for adsorption cooling was presented in 2009 [61].
292 Accordingly, the study of composite adsorbents [62–64] was boosted about 20 years ago [65] with
293 the aim to improve the heat/mass transfer performance by the adsorbents [63,64,66,67]. With the
294 combination of some porous material and the chemical adsorbent yields, these adsorbents such as
295 carbon fiber, activated carbon, graphite are commonly employed [63,68]. Similarly, a study was
296 conducted for the investigation of water vapor adsorption capacity onto activated carbon fiber
297 (ACF), activated carbon powder (ACP), and silica-gel [69]. As a result, the silica gel found an
298 appropriate adsorbent for desalination [69].

299 Consequently, the thermophysical properties of commonly used adsorbents are presented in
 300 Table 1. Extensive studies have been carried out to investigate adsorption isotherms of various
 301 pairs [22,41,48,59,70–79]. The fundamental equations of adsorption isotherm models are
 302 summarized as presented in Table A1. It can be observed in the literature that there is a distinction
 303 in the adsorption mechanisms exhibited by different isotherm models. The possible adsorption
 304 mechanisms are unknown, and thereby the isotherms aids in the determination of adsorption
 305 mechanisms. Consequently, this study deals with the theoretical and experimental understanding
 306 of adsorption desalination technologies concerning performance parameters, comparative analysis,
 307 economic aspects, and future challenges.

308
 309 **Table 1.** The thermophysical properties of commonly used adsorbents in the literature for
 310 adsorption desalination.

Working Pair	BET surface area (m ² /g)	Pore volume (m ³ /kg)	Pore diameter (nm)	Maximum capacity (kg/kg)	Reference
Silica gel 2560/water	636.4	3.27×10^{-4}	1.32	0.32	[70]
Silica gel RD/water	838	4.0×10^{-4}	2.20	0.30	[80]
Fuji silica gel 2060/water	707	3.4×10^{-4}	1.92	0.37	[39]
Fuji silica gel RD/water	780	4.4×10^{-4}	2.24	0.48	[39]
Silica gel A ⁺⁺ /water	863.6	4.89×10^{-4}	1.38	0.48	[70]
AQSOA- Z05/water	187.1	0.7×10^{-4}	1.176	0.22	[81]
AQSOA- Z01/water	189.6	0.712×10^{-4}	1.178	0.215	[81]
Zeolite/water	643	3.1×10^{-4}	1.78	0.25	[82]
AQSOA-	717.8	2.69×10^{-4}	1.184	0.29	[81]

Z02/Water					
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311

312

313 **2. Performance of adsorption desalination system (ADS)**

314 For achieving higher desalination rates and less energy consumption, conventional adsorption
 315 and desalination systems can be modified by applying mass and heat energy processes, improving
 316 the significant components, and compiling renewable energy sources. There are three indicators
 317 to describe the system performance, which indicate the cycle's productivity in cooling,
 318 desalination, and both. The specific daily water production (SDWP) and performance ratio (PR)
 319 are the parameters used for the calculation of the desalination coefficient of performance (COP)
 320 and specific cooling power (SCP) in the case of cooling. The overall conversion ratio (OCR) is the
 321 ratio between useful effects produced for the overall cycle system performance. Third, the
 322 summation of the heat of evaporation and heat of condensation over the input is the heat of
 323 desorption [48,79]. These parameters are presented in the equations (1-5) as given by [40] [73]:

$$COP = \int_0^{t^{cycle}} \frac{Q_{evap}\tau}{Q_{des}} dt \tag{1}$$

$$SDWP = \int_0^{t^{cycle}} \frac{Q_{cond}\tau}{h_{fg}(T_{cond})M_{sg}} dt \tag{2}$$

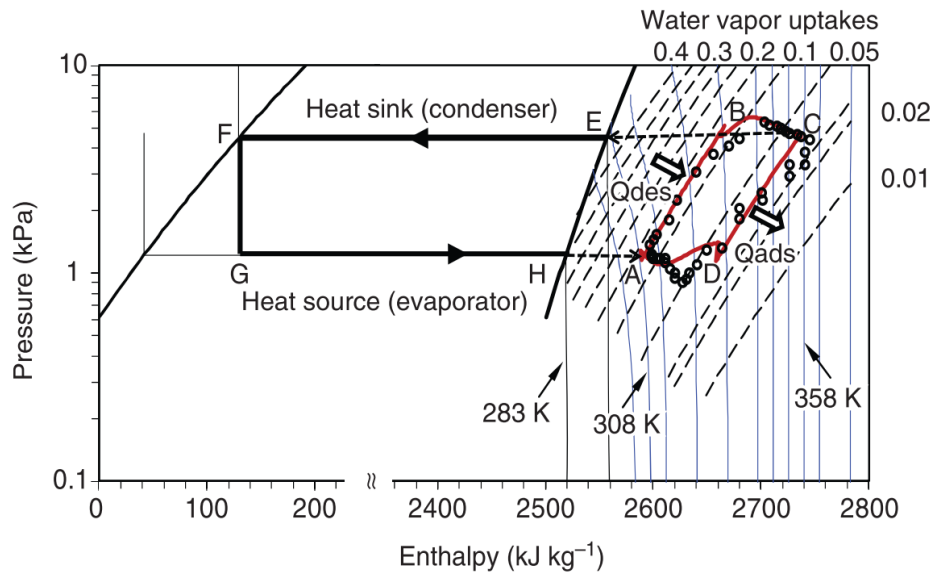
$$SCP = \int_0^{t^{cycle}} \frac{Q_{evap}\tau}{M_{sg}} dt \tag{3}$$

$$PR = \int_0^{t^{cycle}} \frac{\dot{m}_d h_{fg}(T_{cond})\tau}{(Q_{des})} dt \tag{4}$$

$$OCR = \int_0^{t^{cycle}} \frac{Q_{evap} + Q_{cond}}{Q_{des}} dt \tag{5}$$

324 where Q_{evap} , Q_{des} , Q_{cond} , cycle time (s), and time (s) are shown in the nomenclature. Similarly, h_{fg}
 325 is the average radius of adsorption particles (m), M_{sg} is the mass of seawater, T_{cond} is the
 326 temperature of the condenser, and \dot{m}_d is the desalination rate. The system performance improves,
 327 and adsorption of more adsorbent results as adsorption pressure increases. The heat recovery from
 328 beds to evaporator and condenser also results in increased evaporator and condenser profiles. The

329 performance parameters drawn by theoretical and/or experimental study undergo the system
 330 evaluation without considering system energy loss by the components or process.



331
 332 **Fig. 6.** Graphical representation of pressure (P) versus enthalpy (h) for adsorption desalination
 333 cycle [83].

334
 335 It is relatively viable to determine the energetic performances of adsorption cycles from Fig. 6
 336 regarding water production and cooling capacity. The enthalpy of evaporation h_{fg} ($h_g - h_f$) evolves
 337 at the evaporator due to the cooling load (Q_{evap}), and the silica gel surfaces of the sorption bed
 338 absorb the evaporated water vapor [83]. The adsorption cooling systems were analyzed based on
 339 exergy losses due to the thermodynamic parameters reported by Ngoc Vi Cao and Jae Dong Chung
 340 [84]. The CFD used yielded the results of energy analysis. Based on the comprehensive
 341 information obtained from CFD results, the exergy analysis was carried out. The results revealed
 342 that increased energy performance is obtained when the temperature of the heat source increases.

343 Consequently, exergy analysis can be conducted for the complete performance evaluation.
 344 However, the energy available is rudimentary for estimating natural resource utilization, process
 345 economics, and environmental impacts based on environmental conditions [85,86]. Therefore, the
 346 determination of exergy and exergetic efficiency can be calculated from the equations (6) and (7)
 347 as given by [87][88] as:

$$e_f = (h - h^*) - T_o(s - s^*) + \sum_{i=1}^n w_i (\mu_i^* - \mu_i^o) \quad (6)$$

$$E_{etha} = \dot{Q}_{evap} \left(1 - \frac{T_o}{T_{evap}}\right) / \dot{Q}_{des} \left(\frac{T_o}{T_{des}}\right) \quad (7)$$

348 The physical exergy is denoted by the first two terms, whereas the third term represents the
 349 chemical exergy. The temperature (T_o), the concentration of the environment (w_o) and pressure
 350 (P_o) can be referred to as global dead state as the system properties are expressed with ‘0’
 351 subscript. Only the pressure and temperature are varied related to environmental values in the
 352 restricted dead state, denoted by the ‘*’ symbol. The exergy balance equation (8) for the control
 353 volume is given by [87]:

$$\frac{dE}{dt} = \sum \left(1 - \frac{T_o}{T}\right) Q_j + (W_{c.v} - P_o \frac{dV_{cv}}{dt}) + \sum \dot{m}_i e_i - \sum \dot{m}_e e_e - \dot{E}_D \quad (8)$$

354 Exergy analysis aids with the necessary parameters required to complete the design evaluation.
 355 However, it is rudimentary to investigate the exergy destruction effect of the system components
 356 on the operating costs. The thermo-economic variables are a function of thermodynamic
 357 irreversibilities and investment costs [87]. The thermo-economic model for the system’s cost
 358 balancing is based on the governing equations (9.a), (9.b), and (9.c) as given by [89]:

$$\sum_{j=1}^n (c_j \dot{E}_j)_{k,in} + \dot{Z}_k^{CI} + \dot{Z}_k^{OM} = \sum_{j=1}^m (c_j \dot{E}_j)_{k,out} \quad (9.a)$$

$$\dot{Z}_k^{CI} = \frac{CC_L}{\tau} \frac{PEC_k}{\sum_k PEC_k} \quad (9.b)$$

$$\dot{Z}_k^{OM} = \frac{OMC_L}{\tau} \frac{PEC_k}{\sum_k PEC_k} \quad (9.c)$$

359 where τ denotes the total annual time of system operation (hours), \dot{Z}_k is the capital investment cost
 360 of component k. Capital recovery factor (CRF) can be calculated from equation (10) as given by
 361 [87]:

$$CRF = \frac{i_{eff}(1 + i_{eff})^n}{(1 + i_{eff})^n - 1} \quad (10)$$

362 The cost rate associated with exergy loss or levelized carrying charges are determined from
 363 equation (11) as given by [87]:

$$CC_L = TRR_L - FC_L - OMC_L \quad (11)$$

364 Overall, it is observed that lower cost is obtained by achieving higher SWDP, PR, COP, SCP, and
 365 exergy efficiency. A similar condition can be attained by optimizing the operating components
 366 and designing the high-efficiency components.

367

368 **3. Comparative analysis of ADS**

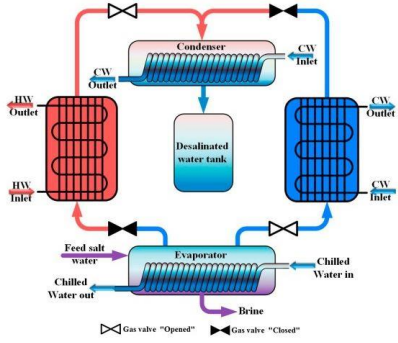
369 Table 2 represents a summarized form of the reported literature regarding the performance
370 parameters of various AD cycles. In addition, the unit features are also represented, such as E_a
371 (kJ/mol), D_{so} (m^2/s), T_{source} , and the unit configuration. The table exhibits the improvement in the
372 performance of traditional AD systems that can be made by employing different techniques such
373 as multi-stage and multi-effect processes along with integrated ADS and similar technologies.
374 Furthermore, as per available literature, the system performance can be improved by applying
375 various modifications in the thermodynamic balance of the system. The key findings from Table
376 2 are presented here. In addition, the dynamic behavior of the adsorption chiller was studied [90]
377 to analyze the effect of adsorbent layers and size on the performance. It has been observed that the
378 smaller layer and grain size improve the coefficient of performance and specific cooling power of
379 the system compared to the large layer and grain-sized adsorption chillers.

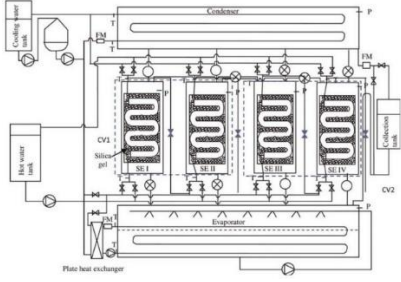
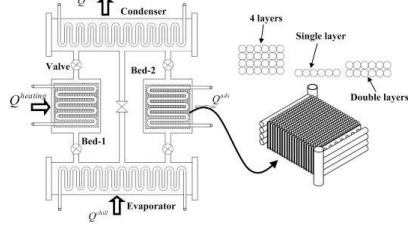
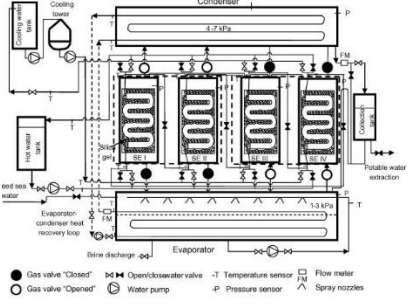
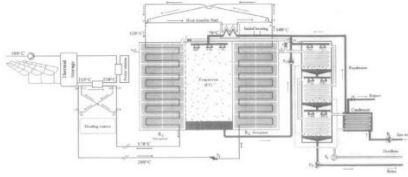
380 Reducing the size of silica gel and layers can increase the system size and cost, but an
381 adsorption bed with high packing density and thermal conductivity can commercialize the chiller
382 [90]. Therefore, a comparative study was conducted for the performance evaluation of a four-bed
383 adsorption cooling and desalination [91]. The adsorbate pairs studied were AQSOA-Z02 + water
384 and silica gel + water as working pairs. It was observed that silica gel is recommended for
385 desalination, and AQSOA-Z02 is suitable for cooling. In addition, a study to analyze the
386 performance of MED and AD hybrid systems was carried out [92]. It was concluded that
387 hybridization improves water production by three folds.

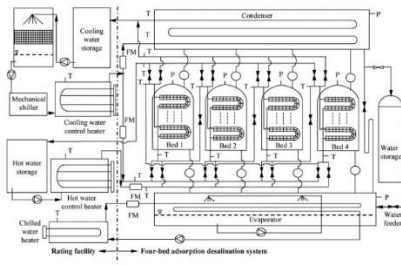
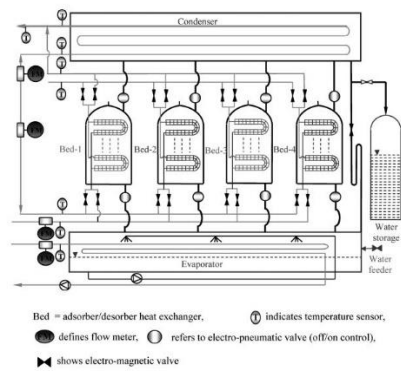
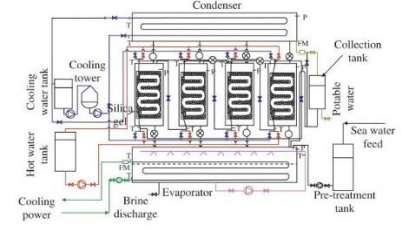
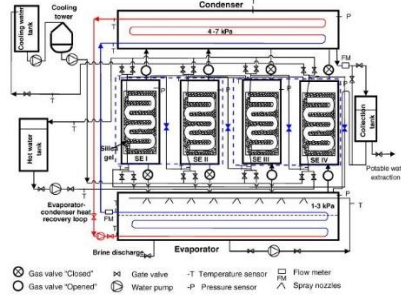
388 Similarly, Wu et al. [100] carried out a thermodynamic study that concluded that lower cooling
389 water temperature improves the system performance. The performance evaluation of solar hot
390 water, waste heat-assisted laboratory-scale 2-bed adsorption cooling cum desalination cycle was
391 performed [93]. It was concluded that achieving a high efficiency from an adsorption cycle
392 produces two advantageous effects: desalting and cooling. In the AD cycle, the waste-heat
393 recovery and conversion for utilization directly impact global warming and carbon emissions. The
394 performance of an advanced adsorption desalination cycle was predicted and modeled [94]. A
395 condenser–evaporator heat recovery scheme was employed, which resulted in a higher water vapor
396 uptake than the traditional cycle, and the performance improvements are evident when plotted on
397 a P–T–C state diagram. A remarkable improvement is observed in the specific water production

398 capacity, which in turn increases the yield. Similarly, when low-temperature waste heat as thermal
 399 energy is used as input, the specific electricity consumption of the advanced cycle is only two
 400 times the thermodynamic limit needed for the desalination of seawater. In summary, the AD
 401 system performance can be improved by employing multi-stage and multi-effect designs, ejector
 402 integrated technology, and operation with various hybrid technologies, driven by solar energy or
 403 using low-grade waste heat.

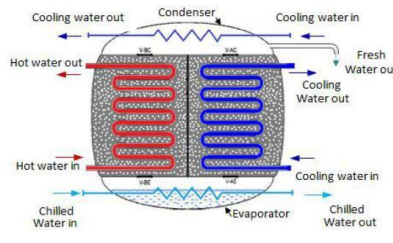
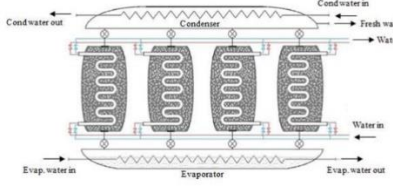
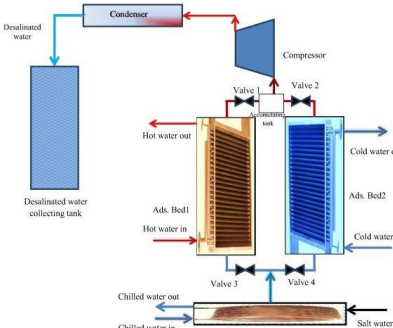
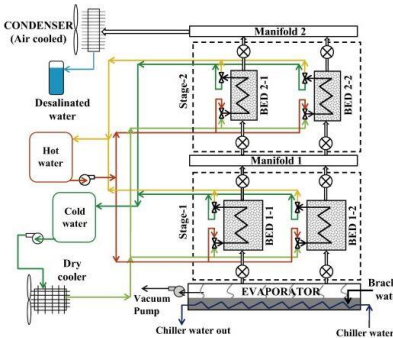
404
 405 **Table 2.** A quick literature review for the performance parameters of various AD systems from
 406 some of the recent studies.

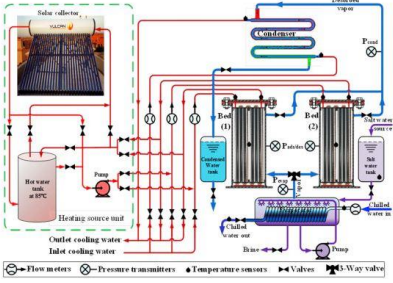
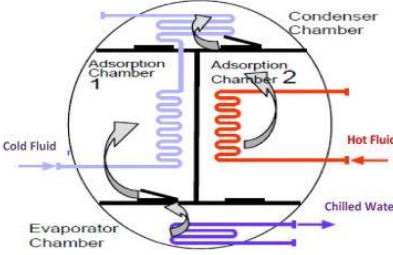
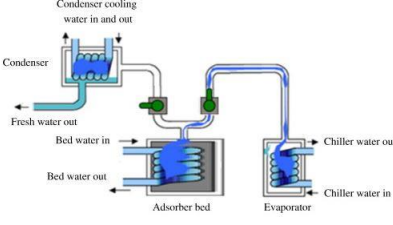
Unit feature and system configuration	T_{source} (°C)	SDWP (kg/kg adsorbent/day)	SCP (TR/ton of adsorbent)	Cycle time (s)	E_a (kJ/mol)	D_{so} (m ² /s)	Schematics	Reference
ADCS employing copper sulfate, driven by low-grade heat sources. Sun-Chakraborty (S-C) and Dubinin-Astakhov (D-A) models have been used for fitting isotherms results, while the linear driving force (LDF) model has been used for the kinetics results. $C = 0.51 \text{ kg/kg}$	25	8.2	64.54	NA	25.053	1.89×10^{-7}		[95]

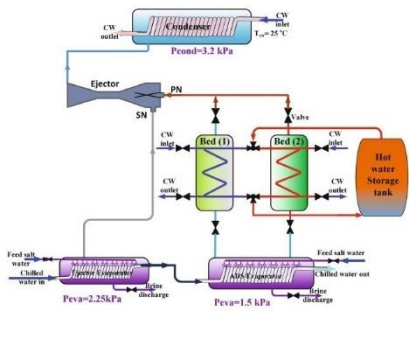
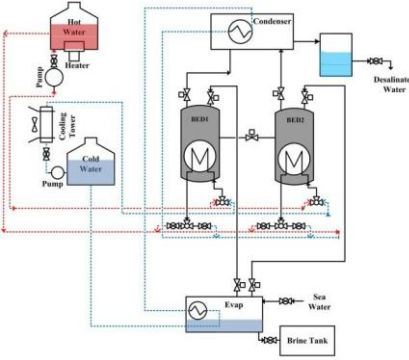
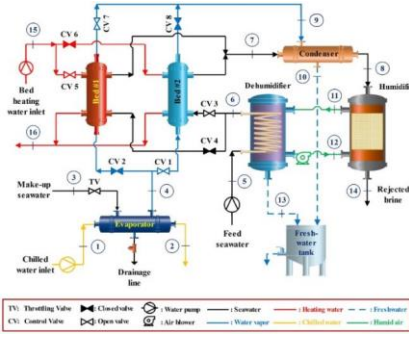
<p>Solar-assisted adsorption (AD) cycle produces two valuable effects, namely cooling and desalination, low-temperature heat input such as thermal energy from solar collectors</p>	7-10	3-5	25 – 35	NA	NA	NA		[93]
<p>Adsorption chiller, single effect, employing two adsorbent beds with various layers of loose grain configurations and silica gel particle sizes, based on experimentally confirmed adsorption isotherms and kinetics data</p>	80	NA	NA	NA	4.2×10^4	2.54×10^{-4}		[90]
<p>Advanced Adsorption desalination (AD) cycle with internal heat recovery between the condenser and the evaporator, investigation on the efficacy of a silica gel-water-based ADC</p>	70	9.34	NA	600	4.2×10^4	2.54×10^{-4}		[47]
<p>MEDAD cycle, adsorption heat pump, open cycle, Zeolite 13X as the solid vapor adsorbent</p>	120–195	0.12	NA	NA	NA	NA		[14]

<p>AD plant, experimental investigation presented concerning assorted primary coolant and feed conditions four beds, single-stage</p>	85	4.7	NA	180	NA	NA		[13]
<p>AD plant, four beds, single-stage, thermally driven cycle</p>	84	7.8	NA	NA	NA	NA		[96]
<p>AD cycle, four beds, single-stage, waste heat-driven, employing 30°C chilled water temperature</p>	85	8	52	960	4.2×10^4	2.54×10^{-4}		[79]
<p>AD plant, four beds, single-stage, silica gel adsorbent</p>	85	12.5	NA	NA	4.2×10^{-7}	2.54×10^{-4}		[59]
<p>AD cycle, advanced two bed with internal heat recovery, recovering the latent heat of condenser and dumping it into the evaporative process of</p>	85	13.46	NA	1440	NA	NA	NA	[49]

the evaporator, silica gel A ⁺⁺								
AD cycle, two beds with internal heat recovery and encapsulated evaporator-condenser, low-grade waste heat driven	85	26	NA	600	4.2×10^{-7}	2.54×10^{-4}	<p>GV-gas valve, AV-electropneumatic valve, P-pump</p>	[94]
MEDAD cycle, vapor uptake by the adsorbent in AD cycle, extracting from the vapor emanating from the last effect of MED, two beds, 10 kg of silica gel	50	5	NA	NA	4.2×10^{-7}	2.54×10^{-4}		[92]
AD system incorporates transient heat and mass transfer processes in the stream-wise direction of the adsorbent bed, two beds, silica gel	80	0.315	NA	NA	42	2.54×10^{-4}	<p> — Hot water supply — Cooling/ Ambient water supply - - - Hot water return - - - Cooling/ Ambient water return → Vapour Adsorption Flow → Vapour Desorption Flow V- Gas Valve, WV- Water Valve, TS- Temperature Sensors PT- Pressure Transducer, VP- Vacuum Pump </p>	[97]

<p>AD cycle, investigation of condenser evaporator on cycle performance, two beds, $T_{\text{evap}} = 30^{\circ}\text{C}$, $T_{\text{cond}} = 10^{\circ}\text{C}$</p>	85	10	77	425	NA	NA		[98]
<p>AD cycle, two adsorbents; silica-gel and AQSOA-Z02 (advanced zeolite)</p> <p>$T_{\text{evap1}} = 10^{\circ}\text{C}$</p> <p>$T_{\text{evap2}} = 30^{\circ}\text{C}$</p>	85	6.2	53.7	600	4.2×10^4	2.54×10^{-4}		[91]
<p>MVC-AD system investigation, two beds, single-stage, a mathematical model has been designed to simulate the operating of the proposed system</p>	NA	14	59.71	NA	41.94	2.54×10^{-4}		[36]
<p>AD cycle, two-stage, two bed, an attempt to model the inter-stage pressure dynamics, $P_{\text{evap}}=1.7 \text{ kPa}$, silica gel RD adsorbent</p>	85	0.9	6.8	3800	NA	NA		[99]

<p>Solar adsorption desalination-cooling (ADC) system, adsorption characteristics of the silica gel-water pair are evaluated, a theoretical dynamic model is developed to predict the system performance, single-stage</p>	95-75	4	33	650	0.51597	2.54×10^{-10}		[100]
<p>AD system, thermodynamic performance evaluation, MOFs suitability assessment for AD, CPO-27 (Ni), single-stage</p>	150	4.3	35.3	700	0.45	NA		[101]
<p>AD system, thermodynamic performance evaluation, MOFs suitability assessment for AD, aluminum fumarate, single-stage</p>	150	6.5	22	700	0.324	NA	NA	[101]
<p>AD system, numerical and experimental investigation of CPO-27(Ni) employment, one bed at $T_{cond} = 5^{\circ}C$ and $T_{evap} = 40^{\circ}C$,</p>	95	22.8	65	600	0.648	NA		[102]

<p>ADEJ-HR cycle, ejector integrated, single cycle to increase the productivity of desalinated water, heat recovery between the condenser and the evaporator</p>	95	40	65	800	32	NA		[103]
<p>AD system, a thermodynamic mathematical model was derived and evaluated by valid experimental data, heat and mass recovery, two bed, single-stage</p>	95	16 - 40	NA	600	0.756	2.54×10^{-4}		[40]
<p>Novel hybrid AD- (water-heated) HDH system, low-grade energy source, investigation of freshwater and cooling capacity, silica gel adsorbent</p>	< 85	9.6	24 - 25	NA	42	2.54×10^{-4}		[104]

407 Key: NA: not available.

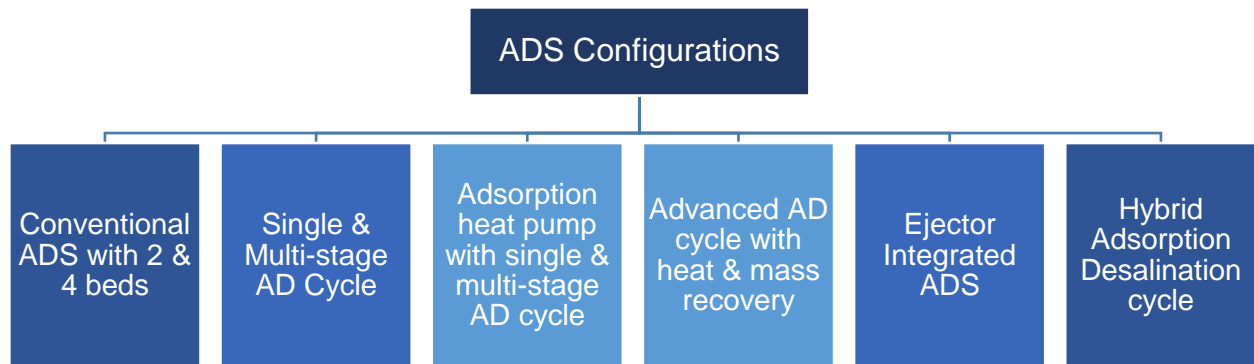
408

409 4. Adsorption desalination (AD) technologies

410 The conventional ADS was developed in 1984 by Broughton [33] derived from adsorption
411 chillers [105–108]. Earlier, freshwater was used as a refrigerant [55]. However, it gives low system
412 performance in terms of COP, cooling capacity [77] with large energy consumption [59].
413 Therefore, the system performance evaluation, along with several modifications, has been
414 presented in the literature. In this regard, various studies have been reported to improve the system
415 performance using appropriate adsorbents with promising adsorption capacity and kinetics [77].

416 Various configurations of adsorption desalination cycles are presented in the literature, which are
 417 listed in Fig. 7. Similarly, a two-stage ejector ADS cycle was proposed [109] to achieve high
 418 efficiency and water production. Hybridization between the AD cycle and two ejectors was
 419 reported along with silica gel-water as an allowing pair to increase the significant productivity.
 420 The hybrid cycle consists of a pair of condensers, evaporators, ejectors, and sorption beds. The
 421 adsorption desalination/2 ejectors system (AD2EJ) was based on COP and SDWP, where
 422 mathematical models for ADS cycle and ejectors are validated. At a regeneration temperature of
 423 85°C and the optimal half-cycle time of ~400s, it was observed from the numerical analysis that
 424 the SDWP and COP are found to be 23.0 m³ per ton of silica gel and 1.64, respectively. Thus, the
 425 freshwater production from AD2EJ is three times higher than conventional AD systems. The
 426 system is further enhanced by connecting the evaporator condenser through an internal heat
 427 recovery circuit. The coming subheading will discuss various AD techniques/configurations
 428 accordingly in detail.

429



430

431 **Fig. 7.** Categorization of adsorption desalination system configurations based on the mode of
 432 operation.

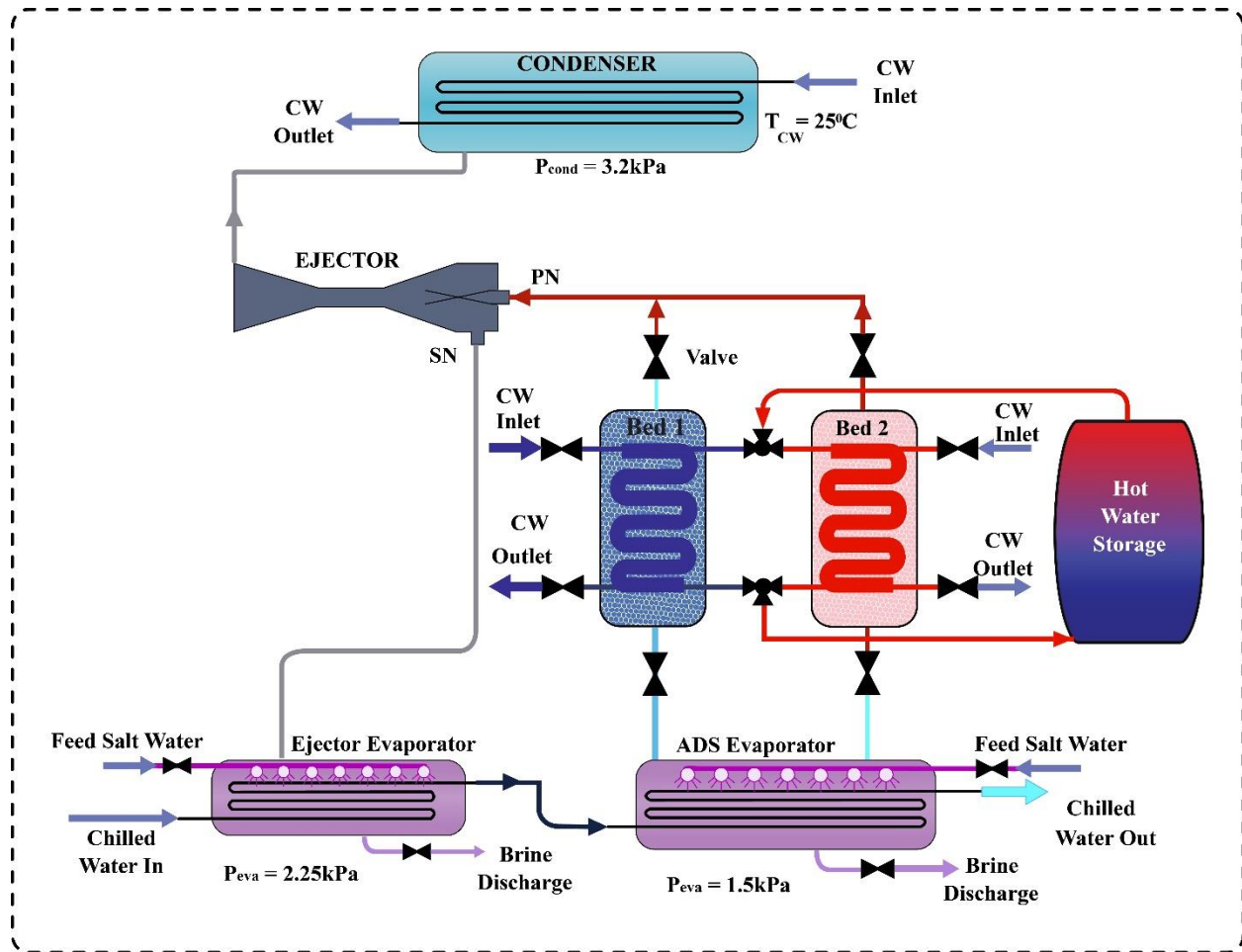
433

434 *4.1. Ejector integrated ADS*

435 A concept of ejector integration with the adsorption desalination cycle has been reported [103].
 436 The concept aimed to increase the productivity and COP of desalinated water. The significant
 437 design improvisation was based on ejector installation after the adsorption bed, which was to be
 438 driven by desorbed vapor. The desalination productivity of the cycle was increased by connecting
 439 the evaporator to the secondary ejector nozzle for vapor extraction [103]. The ejector increases the

440 pressure of secondary vapor, bypassing the primary vapor through a converging-diverging nozzle.
 441 Due to the difference in velocity of secondary and primary flows, the flows do not get mixed until
 442 the sonic velocity of secondary flow is reached [110].

443 An innovative ADEJ cycle is expressed, as shown in Fig. 8 [103]. The presented study involved
 444 the simulation of ADEJ under constraints of uniform pressure and temperatures. Previous literature
 445 has reported the validation and use of the lumped parameters model in multiple studies
 446 [59,95,111].



447
 448 **Fig. 8.** Schematic diagram of integrated ejector adsorption desalination cycle, reproduced from
 449 [103].

450
 451 The study [103] analyzed the effect of an integrated ejector on ADS with condenser/evaporator
 452 heat recovery. The governing mass and energy balance for ADEJ is presented by equations (12)
 453 and (13) as given by [103]:

$$\left[\frac{dM_{SW,evap}}{dt} \right]_{Ejectorevap} = \left[\theta \left(m_{\dot{S}W,in} - \gamma m_b - n \cdot ER \left(\frac{dC_{ADS}}{dt} \right) M_{sg} \right) \right]_{Ejectorevap} \quad (12)$$

$$\left[M_{SW,evap} \cdot \frac{dX_{SW,evap}}{dt} \right]_{Ejectorevap} \quad (13)$$

$$= \left[\theta X_{SW,in} m_{\dot{S}W,in} - \gamma X_{SW,in} m_b - n \cdot ER \cdot X_D \left(\frac{dC_{ADS}}{dt} \right) M_{sg} \right]_{Ejectorevap}$$

454 where n, r, and Θ are switching operating modes [59]. ER is the ejector entrainment ratio equal to
 455 the secondary fluidness flow rate to the primary fluid mass flow rate. The regeneration temperature
 456 has an insignificant effect on the ejector entrainment ratio and ejector compression ratio. The
 457 average ER and CR were about 0.5 and 1.42, respectively. Therefore, many studies investigated
 458 the vapor compression cycle (VCC) to benefit the kinetic energy available to improve the cycle
 459 COP. In a vapor-vapor ejector refrigeration cycle, the ejector functions similar to a compressor.
 460 As inlet secondary vapor passes through a converging-diverging nozzle, the pressure is increased
 461 by the ejector [59].

462 A study had been carried out which investigated the integrated ejector within the vapor
 463 compression cycle. The study was primarily focused on testing the effect of condensing,
 464 generating, and evaporating temperatures of the ejector sub-cycle on the ejector vapor compression
 465 cycle. Another study [112] proposed an ejector with hybrid VCC, which reported that EVCC
 466 performance largely depends on the ESC temperatures. Nevertheless, there is a similarity in the
 467 variation of the degree of sub-cooling at the vapor compression sub-cycle (VCSC) to improvement
 468 in COP at ejector vapor compression cycle over vapor compression sub-cycle. Further, it was
 469 observed that the system under certain operating conditions yields a comparatively more
 470 significant COP improvement of about 15.9-21.0%.

471 A hybrid vapor compression refrigeration system (HVCR), which consists of a combined
 472 ejector refrigeration (ER) system and vapor compression refrigeration system (VCR), was
 473 simulated [113]. It was found that the ejector and gas cooler accounted for a large percentage of
 474 total exergy destruction. Therefore, the HVCR system expressed an improved COP by 25% than
 475 the conventional VCR system. Furthermore, an adsorption refrigeration system was proposed
 476 [114]. The system utilized two ejectors as the cycle power source in the place of mechanical
 477 pumps. The location of the liquid-vapor ejector is at the absorber inlet. It has been concluded that
 478 the novel combined double ejector refrigeration cycle (DEARC) when compared with the
 479 conventional air-cooled absorption refrigeration cycle, which sacrifices a portion of refrigerant,

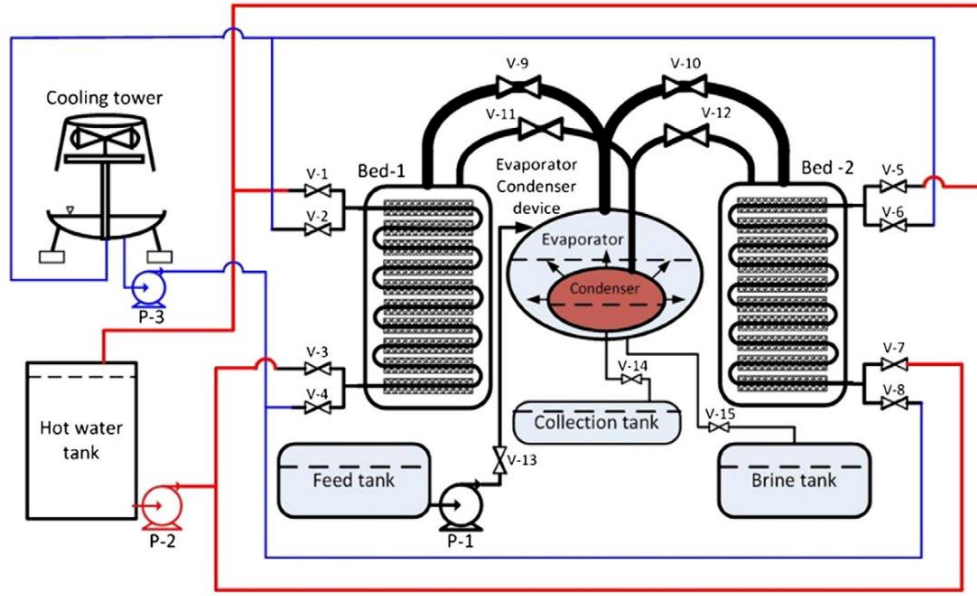
480 leads to a decreased COP. Notwithstanding this, a maximum COP of ~0.63 can be reached which
481 encourages the modification and practical use under exhaust heat conditions.

482 A simple ejector without moving parts combined with a solar-assisted absorption cooling
483 system employing LiBr/H₂O as a working fluid was studied. The system performance evaluation
484 was expressed [115]. As a result, it was observed that the post addition of the condenser along with
485 condenser and evaporator load is found to be perpetually increased as compared to the primary
486 cycle, which suggests that the increase in the system's cooling capacity is due to the increase in
487 entrainment ratio and the evaporator temperature. A 60% increase of COP of the modified cycle
488 was witnessed to that of the primary cycle at the same temperature and pressure. Adopting this
489 process enhances the functions, stabilizes the refrigeration system, and qualifies the system to work
490 under the raised temperature of the condenser. A combined ejector integrated system is believed
491 to work efficiently with other systems and increases the COP. The main goals are the maximization
492 of AD system performance by ejector integration before condenser in an ADS technology.
493 Simulation, experimental and theoretical scenarios contribute to paving the way for researchers to
494 industrialize AD systems. Furthermore, the ADEJ system using energy and mass balance ejector-
495 integration strategy can improve the system performance.

496

497 *4.2. Integrated evaporator-condenser cascaded ADS*

498 An AD plant layout was developed and evaluated using numerical simulation [48,49]. The AD
499 cycle suggested in Fig. 9. uses internal heat recovery and heat transfer between the encapsulated
500 condenser and evaporator. Shell-and-tube heat exchangers are used in the integrated evaporator-
501 condenser unit. One of the benefits of this integration is the employment of working fluid circuits
502 to heat the evaporator and cool the condenser, which leads to a considerable reduction in the cost
503 of pumping power. This arrangement also reduces the resistances of heat transfer and increases
504 seawater evaporation rates. Theoretical findings showed that this advanced arrangement could
505 produce an SDWP of 26 kg/kg of silica gel per day utilizing a hot water temperature of 85 °C,
506 twice as high as the primary plant. Furthermore, the performance of other configurations,
507 comparable to the current pilot adsorption desalination plant at NUS, was statistically investigated.
508 This alternate configuration with internal heat recovery is an entirely integrated condenser-
509 evaporator system that employs a cooling fluid circuit to provide the condenser's condensation
510 heat to the evaporator [116].



511
 512 **Fig. 9.** Schematic diagram of AD cycle with an integrated evaporator-condenser unit, [59].

513 Integrating the condenser-evaporator, both coolant circuits for essential cycle operation are
 514 omitted, resulting in considerable savings in pumping costs. Instead, heat is transmitted directly
 515 through the walls of the condenser tubes within the evaporator shell. As a result, the device's heat
 516 transfer resistances are decreased, which improves the film evaporation of water vapor from
 517 seawater. In addition, a benefit of combining the condenser-evaporator unit is that the evaporator
 518 has a more significant vapor pressure. Therefore, it significantly enhances the rate of silica gel
 519 vapor absorption over the specified adsorption process. The benefits of the advanced AD cycle
 520 design, i.e., an integrated condenser-evaporator [59] are: (i) a decrease in parasitic electrical power
 521 as a result of the exclusion of pumps from the chilled and cooling water circuits, (ii) the
 522 pressurization effect improves the silica gel's adsorption capacity, (iii) the effective condensation
 523 temperature in the condenser is lowered, which facilitates desorption, and (iv) the heat source
 524 temperature is lowered to 50°C for the AD cycle.

525 Several studies have further investigated the condenser-evaporator integrated unit. The unit was
 526 explained by Thu et al. [47,49] for various heat recovery designs of ADS on 2 or 4-bed modes.
 527 The integrated evaporator-condenser was further analyzed for the effective mass of ADS with heat
 528 recovery to obtain the desired results. The evaporator's temperature is raised to ensure condenser
 529 heat recovery, leading to an increase in water production. This amount later improved thrice than
 530 the conventional ADS [15,46,117]. A study was carried out as the comparative analysis of

531 integrated evaporator-condenser and conventional adsorption desalination system. The newly
532 developed system's cycle design consisted of four beds, including a condenser, evaporator, and
533 one integrated evaporator-condenser unit. The cycle is composed of two single-stage regular ADC.
534 The new multi-cycle configuration benefits from reducing condenser temperature, increasing the
535 quantity of desalinated water and generating cooling power at the reduced evaporator temperature.
536 Consequently, the integration of heat and mass aids ADC performance improvement.

537

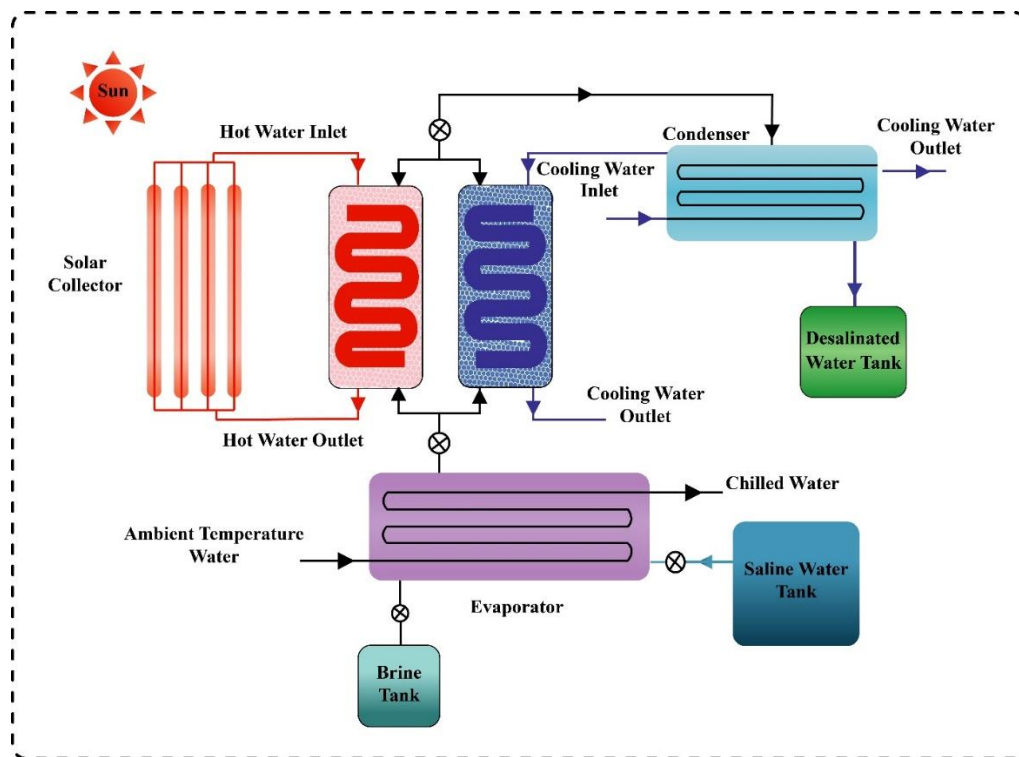
538 *4.3.Solar/sunlight regenerable ADS*

539 Waste heat and solar energy can be efficiently used to run the adsorption desalination cooling
540 system (ADCS) [118]. A 4-bed single-stage solar-driven ADS was developed and analyzed for
541 system dynamics [119]. The system consisted of a non-concentrating solar collector providing
542 low-grade heat, utilized by the thermal desalination cum refrigeration cycle. It was designed to
543 generate cooling and desalinated water when combined thermal compression and flash
544 evaporation are employed. It is observed that a system cooling and desalination capacities are
545 affected with solar energy utilization simultaneously. The condenser temperature rises with the
546 the increase in operating pressure ratio which rapidly degrades the system performance.
547 Consequently, two-stage ADS can be employed which aids in improving system performance.

548 A detailed thermodynamic analysis of the solar-based adsorption desalination and cooling
549 system was performed as shown in Fig. 10 [120]. The system utilizes the low-grade heat from
550 the solar collector to drive the system. The thermal compression and combined flash were
551 employed for the production of cooling and potable water. The findings indicate that the hot
552 water intake temperature, cooling water temperature, and condenser temperature all significantly
553 influence the system's water output, energy consumption, and COP. Low cooling water
554 temperature, low condensation temperature, and high hot water intake temperature result in
555 improved system performance. The rate of water production increases as the temperature of the
556 hot water intake rises, eventually flattening off at higher temperatures. This rising trend of water
557 output with temperature is caused by an improvement in the amount of water desorbed from the
558 adsorbent as the hot water temperature rises.

559 The performance evaluation of solar-driven AD cooling system was conducted [100]. The
560 system was tested with 13.5 kg of commercially available silica gel. The working pair of silica
561 gel-water was evaluated and the system performance prediction was made based on theoretical

562 dynamic model. The system resulted in COP, average specific daily water production, the
563 average specific cooling power of 0.45, 4 m³ per ton of silica gel, and 112 W/kg, respectively. A
564 study on the water quality analysis of the solar-driven adsorption desalination cycle was
565 presented [121]. A significant reduction in TDS levels of feed seawater from approximately
566 40,000 ppm to less than 10 ppm is evidence of process efficiency. In the desalinated water, there
567 are deficient levels of chloride, magnesium, sulfate, silicate, sodium, calcium, and bromide, even
568 less than 0.1 ppm. The conductivity parameters of product water were composed of the distilled
569 water conductivity levels ranging between 2 and 6 μS/cm.
570



571
572 **Fig. 10.** Schematic diagram of a solar-driven adsorption desalination system, reproduced from
573 [120].
574

575 The performance investigation of the solar-driven dual effect adsorption cycle for the
576 production of potable water and cooling effect was performed by Ng et al. [93]. It is driven at low-
577 temperature heat, such as thermal energy provided by solar collectors. Mathematical modeling of
578 the system was discussed along with key parameters based on SCC and SDWP. The system uses
579 a flat plate type solar collector and silica gel-water based adsorption desalination cycle. The system

580 experimentation exhibit capacity of chilled water production at a temperature and SCC range
581 variation of 7-10°C and 25-35 Rton/tonne of silica gel, respectively. Consequently, the adsorption
582 desalination cycle possesses a conversion or performance ratio of 0.8-1.1 and daily production
583 capacity, SDWP of 3-5 m³ per tonne of silica gel per day. Similarly, a solar-assisted adsorption
584 refrigerator was developed [122]. The system performance can be improved by well-insulating the
585 cold chamber and using automatic valves, potentially improving efficiency. A numerical and
586 experimental investigation was carried out by Ahmadi et al. [123]. The feasibility investigation of
587 the solar eductor-assisted desalination system performed resulted in a promising performance of
588 the closed educator installed at the exterior of the evaporation chamber. It was revealed from the
589 experimental analysis that reducing the flow rate of water can reduce energy requirements. A
590 thermodynamic framework was presented by the authors [7] to address energy efficacy. It was
591 observed that there is 2.5-3% energy consumption by the thermally driven system when integrated
592 with power plants. Furthermore, it must have innovative systems to fulfill the 20-30%
593 thermodynamic limit by 2030.

594 Desalination technology is extensively employed for the reduction of escalating water scarcity.
595 Currently, membrane and thermal-based desalination technologies are used to address the issue
596 and freshwater production. Nevertheless, thermal technologies' energy consumption is relatively
597 high. For example, multi-effect and multistage desalination have an energy consumption of 14-27
598 kWh/m³. The drawbacks of membrane-based techniques include the chemical consumption in
599 dechlorination & cleaning and relatively high energy consumption of 1.6-6 kW/h³ [124–126].
600 Similarly, efficient salt removal can be conducted by passing brackish water from the adsorbent
601 or ion-exchangers-filled columns [127]. Mainly, in the water industry, there is an implementation
602 of thermo-responsive ion adsorbent-based desalination. The process of ion-adsorption is energy
603 efficient, but for regeneration, the requirement of thermal energy at increased temperature (i.e.,
604 80°C) is substantial [126,128–130].

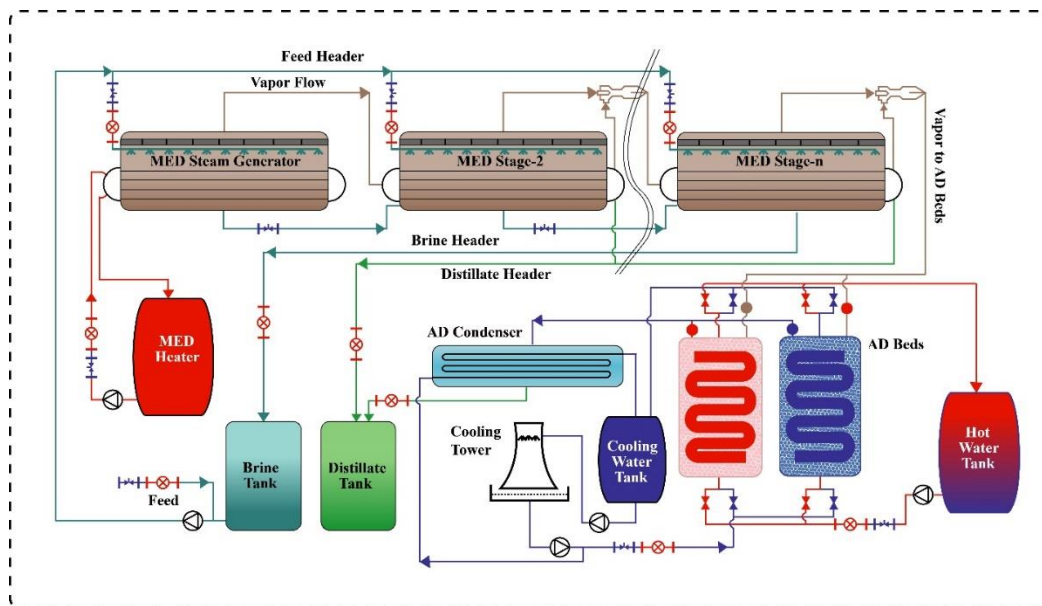
605

606 *4.4. Multi-effect ADS*

607 A parallel feed MED and AD system integrated is usually employed to replicate the process.
608 The schematic diagram is shown in Fig. 11 [92]. The details of the AD cycle function can be
609 studied from the literature [41]. In another study [92], the suggested simulation model by the
610 authors for the MED system, horizontal tubes falling film evaporators are employed. The feed is

611 de-aerated to eliminate the oxygen (through a de-aeration tank with a vacuum pump) before being
 612 fed parallel to all evaporators—built-in pre-heaters route feed-in evaporators before being sprayed
 613 onto a tube bundle through spray nozzles. The first stage, often known as the steam generator, is
 614 utilized to generate initial steam. The hot water is cycled via the tubes, and pressure is controlled
 615 to evaporate the seawater at the appropriate temperature, corresponding to the hot water
 616 temperature. The vapors generated are then allowed to move to the tube side of the subsequent
 617 effect. Condensation heat is utilized to vaporize the feed sprayed at that step. Similarly, vapors are
 618 cascaded in subsequent phases until the final effect is achieved (nth effect). Intermediate steam jet
 619 ejectors are supplied to remove non-condensed vapors from each stage distillate box. The final
 620 stage of MED is linked to AD beds to adsorb the vapors onto adsorbent (silica gel) packaged in
 621 the shape of cakes. Because of the strong affinity of the adsorbent for water vapors, this
 622 combination of MED and AD lowers the temperature of the last stages below ambient, allowing
 623 for more stages and higher recoveries while also reducing the possibilities of corrosion. The vapors
 624 are desorbed and condensed in the condenser with the help of the heat source. A vacuum pump is
 625 introduced to keep the pressure level in the MED stages and the AD machine constant. It also aids
 626 in removing air that has entered the system as a result of a leak. A distillate collecting tank is
 627 supplied to collect distillate across all stages of the MED and the AD condenser.

628



629

630 **Fig. 11.** Schematic representation of Multi-Effect Adsorption Desalination System (MEDAD),
 631 reproduced from [92].

632

633 The thermodynamic synergy can hybridize the emerging adsorption desalination systems and
634 the conventional ADS for their thermodynamic synergy. The authors carried out a similar
635 hybridization of conventional multi-effect distillation (MED) with an emerging low-energy
636 adsorption cycle, which resulted in an advanced desalination cycle [92]. As the adsorbent uptakes
637 the vapor, it yields the cycle integration and extraction from the vapor emanating from the last
638 effect of MED. Additional evaporation-condensation stages can be accommodated by the system
639 as the temperature difference range is increased. The study presented the numerical model of the
640 proposed MEDAD cycle, which comprises the water production rate of the conventional MED
641 cycle. The hybridized MED scavenges heat from the ambient air and allows the latter stages of
642 MED to operate at a lesser temperature than the ambient air. The latter stages' reduced saturation
643 temperatures eradicate the fouling and scaling effects, but improved water recovery from feed
644 seawater may increase solution concentration. Similarly, a study on efficiency examination was
645 carried out by the authors [131]. A MED system was investigated, and it was found that seawater
646 thermocline (ST)-driven MED system can increase the system efficiency 2 folds and be regarded
647 as the most appropriate 'green desalination' method.

648 Some studies were also reported on the thermo-economic analysis of MED systems by
649 employing exergy analysis and optimization methods [132–137]. In addition, various heat
650 performance pump coupled MED systems were also studied by researchers [138,139]. For
651 example, an integrated adsorption vapor compression (ADVC) system was proposed [140]. The
652 system employs the vapor obtained from the last stage of MED for adsorption, whereas, from
653 desorption, the regenerated water vapor is forwarded to the first effect of the MED cycle.
654 Consequently, the MED system condenser is eliminated in the combined cycle, and the
655 performance ratio is improved to two folds [141–144]. Notwithstanding this, some other research
656 on similar technologies, such as vacuum multi-effect membrane distillation technology, was
657 investigated by Chen et al. [145].

658 A hybrid MEDAD cycle was analyzed by the authors [146]. The system's hybridization allows
659 the conventional MED cycle to be operated at a low temperature of 5°C. As a result, the MEDAD
660 cycle resulted in a significant increase in distillate production as the MED can scavenge the
661 ambient energy whereas, the conventional AD cycle is driven by low-temperature waste heat
662 obtained from renewable or exhaust sources. The study's experimental setup consisted of MEDAD

663 and 3-stage MED plants, assorted and tested at 15°C-17°C temperature from a waste heat source.
664 An in-depth observation and monitoring of the system result in the MEDAD cycle's synergetic
665 matching lead to a significant increase in distillate production of about 2.5-3 times compared to
666 the conventional MED cycle.

667 A hybrid MEDAD cycle was developed [173], as shown in Fig. 11, operating at sub-
668 atmospheric temperatures and pressures. A quantum performance enhancement has resulted from
669 the hybridization of the multi-effect distillation and adsorption desalination cycle. The symbiotic
670 enhancement resulted in brine heater usage emanated waste heat, which cascaded the adsorbent's
671 regeneration temperature. Furthermore, the decreased saturation temperature of 5°C was provided
672 by vapor extraction from the last MED stage by AD cycle, resulting in scavenging of heat leaks
673 into the MED stages from the ambient. The hybrid desalination system augments the desalination
674 plant's capacity for water production twice. A simulation of the 8-stage MED cycle was used for
675 the demonstration of a hybrid cycle. The silica gel's sorption properties were employed as a
676 mechanical vapor compressor to reduce the saturation temperatures of MED stages. The kinetics
677 of adsorbent-adsorbate such as silica gel-water and adsorption isotherm were employed for
678 modeling energy, mass, and concentration equations. The MEDAD cycle was operated at 65°C-
679 90°C temperatures, and upon comparison with conventional MED, the cycle exhibited an
680 augmented water production rate from 60% to two folds and an increase in GOR and RP by 40%.

681

682 *4.5.Hybrid ADS*

683 In recent years, various studies have been conducted to address the limitations of conventional
684 ADS by designing some improved designs [147]. The conventional ADS works at a relatively low
685 temperature where evaporation occurs in the ambient, typically around 5°C to 20°C [72]. Solar
686 heating was employed to desorb the water vapor, which condenses at high altitudes and produces
687 pure water without any supply of carbon-based energy. Furthermore, theoretical analysis was
688 performed on a low-grade solar heat-driven water desalination system [11]. The system possessed
689 distinction from conventional ADS as natural forces create vacuum conditions incorporated into a
690 single design consisting of an evaporator, condenser, solar heating systems, and injection. The
691 numerical simulations of the system resulted in 90% or higher efficiency. The simulation for a
692 solar-driven adsorption desalination system was performed [14]. The theoretical model combined
693 with an adsorption heat pump using zeolite was analyzed to determine the system energy

694 consumption and water production. A solar-driven dual effect adsorption cycle was studied by Ng
695 et al. [93]. The solar AD cycle's performance investigation resulted in desalination and cooling at
696 low-temperature heat input, i.e., thermal energy from solar collectors. Further mathematical and
697 experimental modeling resulted in a performance ratio of 0.8-1.1 with producing SDWP of 3-5 m³
698 per tonne of silica gel per day at a temperature of 7-10 °C. Consequently, a low-grade heat-driven
699 adsorbent cycle integrated with membrane processes can ensure 150% of chemical rejection and
700 ensure about 99% energy-saving [148]. Furthermore, as renewable energies are environment
701 friendly having no negative impact, the desalination systems are driven by renewable technologies
702 to ensure the future sustainability of water supplies [149].

703 An innovative pressure swing adsorption cycle (PSAD) was proposed by the authors [150], which
704 employs a thermal vapor compressor for steam regeneration. The study revealed that primary
705 steam having 2 bar of pressure can generate a compression ratio of 3-4. Consequently, the
706 desalination cycle can be operated by re-utilizing the discharged steam, hence maximizing the
707 exergy of steam. Several other studies were performed for the analysis of AD cycle at different
708 regeneration temperatures examining the operational strategy, theoretical model, numerical
709 simulation, and performances at different heat source temperatures [151][152] [44]. Consequently,
710 a similar study was also reported [153]. To analyze the performance of desalination and cooling
711 systems concerning time, a study investigated two-half cycle ADS [154]. The effect of time
712 variation by changing the interval of adsorption-desorption by 200s to 700s and switching interval
713 of 20-40s. At cooling water inlet, chilled water inlet and hot water temperature of 35°C, 7°C, and
714 85°C, respectively. The analysis resulted in the SCP, SDWP as 25TR, and 12 m³ per ton of silica
715 gel, respectively. A silica gel-water, 4-bed adsorption desalination system was studied [119]. The
716 performance evaluation of various condenser temperatures and cycle time was carried out for the
717 determination of optimum requirements for desalination and cooling. The simulations pointed out
718 that the time range of the single-stage adsorption desalination system has an optimum half-cycle
719 time of 600 to 900s to increase the cooling and desalination capacity. Therefore, as the cycle time
720 is increased, the coefficient of performance is raised.

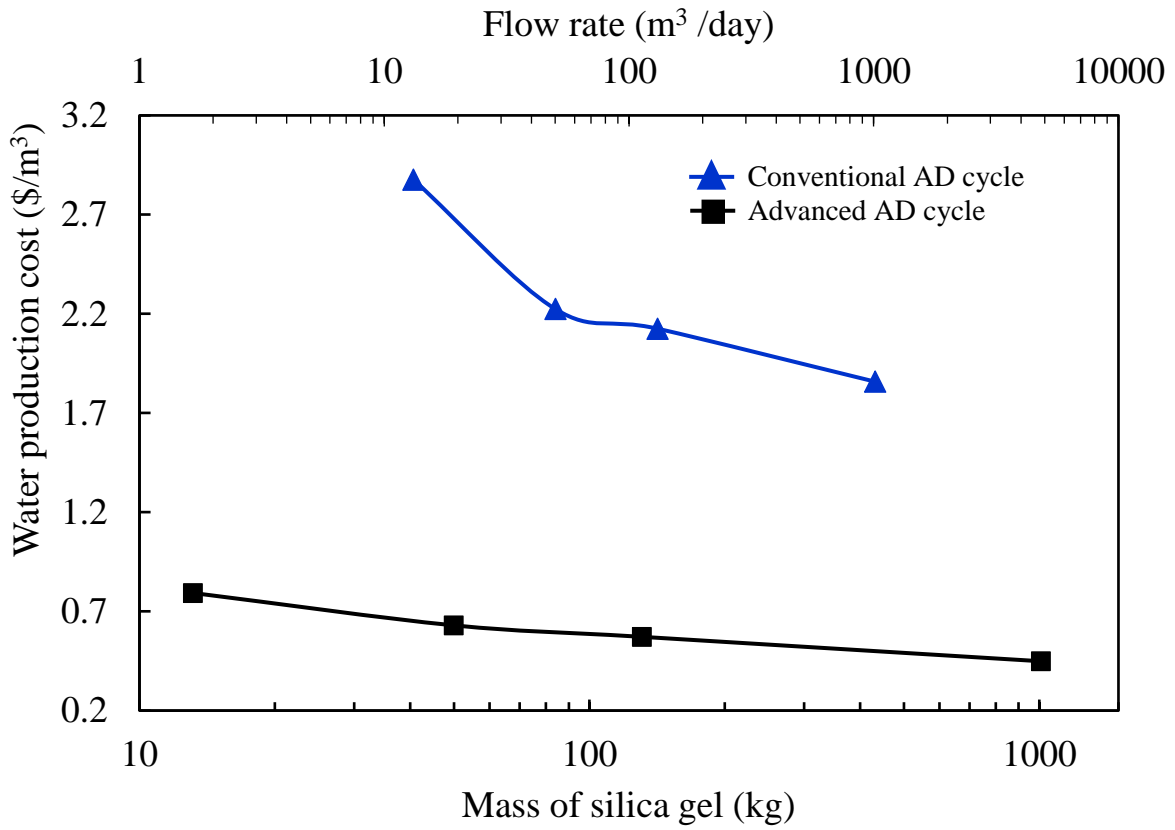
721 The studies discussed have shown the performance of silica gel as an adsorbent that can be
722 utilized in the low heat-driven ADS, usually at a temperature of 85 °C. Various researches were
723 conducted by Kim et al. [95] to evaluate ferroaluminophosphate adsorbent for its potential use in
724 ADS. FAM-Z01 possesses an equilibrium adsorption capacity 5 times higher than the silica gel

725 adsorbent at a temperature of 40 °C. The performance of the adsorption desalination cooling cycle
726 and the subsequent effect of evaporator-condenser temperature were mathematically studied.
727 Further investigation revealed that when the condenser temperature is 30 °C, and the water inlet
728 temperature is 10 °C, the system results in SCP and SDWP as 77 TR per ton silica gel and 10 m³
729 per ton of silica gel, respectively.

730

731 **5. Economic aspects**

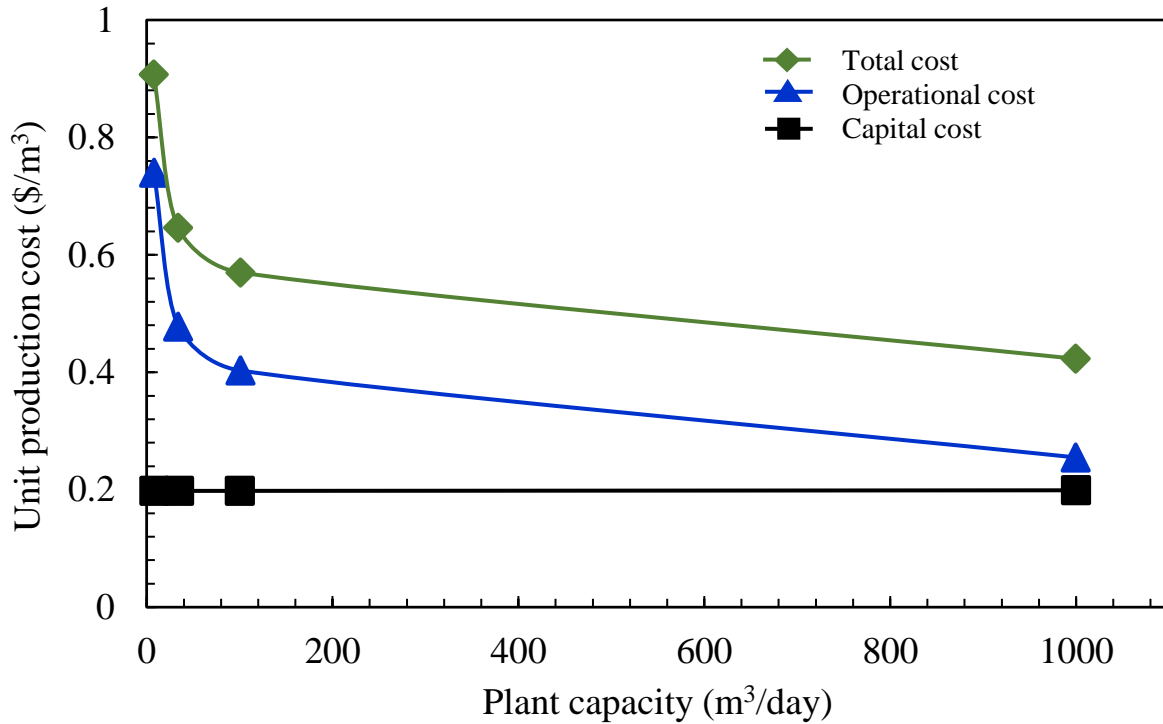
732 The advanced and conventional adsorption desalination plants of different capacities/ flow
733 rates were analyzed for water production cost as a function of the mass of adsorbent, as shown in
734 Fig. 12 [155]. It has been observed that the unit production cost of a conventional adsorption
735 desalination system is nearly four times higher (i.e., ~\$1.91/m³) than the advanced cycle (i.e.,
736 ~\$0.457/m³). Factors affecting the unit cost of the AD system are (i) reduction in pumping power,
737 (ii) design improvement, and (iii) improvement in the overall heat transfer coefficient of
738 condenser-evaporator integrated design. An AD plant of 1000 m³/day capacity has the unit
739 production cost (operational cost+capital cost) of \$0.457/m³, as shown in Fig. 13. The upscaling
740 of AD plants causes a reduction in capital cost [155]. The specific energy consumption and the
741 energy cost of water production of the AD cycle are compared with other conventional methods
742 such as RO, MSF, and MED [96]. The MSF cycle shows the highest energy cost of water
743 production (i.e., US \$0.647 per m³), whereas the AD cycle gives the lowest energy consumption
744 of about 1.5 kW/m³, which is equivalent to the production cost US \$0.227 per m³. It has been
745 concluded that the AD technology can desalinate the high salinity feed water and produce low
746 salinity water at a low cost, i.e., US \$0.2/m³ [96].



747

748 **Fig. 12.** Relationship between the cost of water production and the mass of silica gel

749 reproduced from [155].



750
 751 **Fig. 13.** The relationship between the unit production cost and plant capacity of an AD plant,
 752 reproduced from [155].

753
 754 **6. Conclusions**

755 Adsorption desalination (AD) is a emerging technique for obtaining potable water from an
 756 unlimited supply of brackish water. This study discusses adsorption desalination (AD) systems
 757 and subsequent hybridization options. The AD systems are reviewed for energy consumption,
 758 water productivity, and associated performance parameters. The AD system performance has been
 759 improved with the integration of multi-effect AD designs (MED), ejector integrated technology,
 760 evaporator-condenser cascaded technology, solar and hybrid options. For instance, a combined
 761 ejector technology can stabilize the refrigeration system and increase the COP of the desalination
 762 plant. Similarly, the evaporator-condenser AD systems reduce condenser temperature, increase
 763 desalinated water production, and generate cooling power. The specific energy consumption of
 764 ADS is <math><1.5 \text{ kWh/m}^3</math>. For solar AD systems, the specific cooling capacity is 112 W/kg, and the
 765 COP is 0.45. The SDWP was observed as 66% for a heat and mass recovery process of 2-bed ADS.
 766 The cost of an advanced AD cycle is $\sim \$0.457/\text{m}^3$ which is four times less than a conventional AD

767 cycle. Compared to standard MED plants, the hybridization of MED+AD improves output more
 768 than 3 times with the same TBT. The AD cost is still a challenge which can be improved with the
 769 further development and research for SDWP enhancement at low heat source temperature. The
 770 AD is emerging as a viable and realistic option for relieving global thirst and thereby going to be
 771 accessible in the global market.

772 **CRedit authorship contribution statement**

773 **Nadia Riaz:** Conceptualization, Methodology, Software, Formal analysis, Investigation,
 774 Writing - Original Draft. **Muhammad Sultan:** Conceptualization, Methodology, Validation,
 775 Resources, Writing - Original Draft, Visualization, Supervision, Project administration, Funding
 776 acquisition. **Takahiko Miyazaki:** Validation, Resources, Writing - Review & Editing,
 777 Supervision, Project administration, Funding acquisition. **Muhammad W. Shahzad:** Formal
 778 analysis, Software, Writing - Review & Editing, Visualization. **Muhammad Farooq:** Data
 779 Curation, Writing - Review & Editing, Visualization. **Uzair Sajjad:** Writing - Review & Editing,
 780 Visualization. **Yasir Niaz:** Writing - Review & Editing, Visualization.

781

782 **Declaration of Competing Interest**

783 The authors declare that they have no known competing financial interests or personal
 784 relationships that could have appeared to influence the work reported in this paper.

785

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790

791 **Appendix A**

792 **Table A1.** Mathematical expressions for the classification of adsorption isotherms [156].

Classification	Name of the model	Equation of the model	Reference
Empirical adsorption isotherm models	Linear isotherm model (Henry's law)	$q_e = K C_e$	[156]
	Freundlich isotherm	$q_e = K_F C_e^{1/n}$	[157]

	model		
	Redliche-Peterson (Re-P) isotherm model	$q_e = \frac{K_{RP}C_e}{1 + a_{RP}C_e^g}$	[158]
	Sips isotherm model	$q_e = \frac{q_{ms}K_S C_e^{n_s}}{1 + K_S C_e^{n_s}}$	[159]
	Toth isotherm model	$q_e = \frac{K_T C_e}{(a_T + C_e^Z)^{1/z}}$	[160]
	Temkin isotherm model	$q_e = \frac{RT}{b} \ln(AC_e)$	[161]
Adsorption models based on Polanyi's potential theory	Dubinin-Radushkevich (D-R) isotherm model	$q_e = q_{mD-R} e^{-K_{DR} \varepsilon^2}$ $\varepsilon = RT \ln \frac{C_S}{C_e}$	[162]
	Dubinin-Astakhov (D-A) isotherm model	$q_e = q_{mD-A} e^{[-(\frac{\varepsilon}{E_{DA}})^{n_{DA}}]}$ $\varepsilon = RT \ln \frac{C_S}{C_e}$	[162]
Chemical adsorption models	Langmuir isotherm model	$q_e = \frac{q_m K_L C_e}{1 + K_L C_e}$	[163]
	Volmer isotherm model	$b_V C_e = \frac{q_e}{q_{mV} - q_e} e^{\frac{q_e}{q_{mV} - q_e}}$	[164]
Physical adsorption models	BET isotherm model ($n \geq 1/4 \infty$)	$q_e = \frac{q_{mBET} K_{BET} C_e}{(1 - K_{BET} C_e)[1 - K_{BET} C_e + K_{BET} C_e]}$	[165]
	Aranovich isotherm model	$q_e = \frac{q_{mA} C_A \frac{C_e}{C_{SA}}}{\sqrt{\left(1 - \frac{C_e}{C_{SA}}\right) \left(1 + C_A \frac{C_e}{C_{SA}}\right)}}$	[166]
Ion exchange isotherm model	The homovalent ion exchange model	$\frac{c_A}{q_A} = \frac{1}{\xi} \left(c_A + \frac{c_B}{K_{B,A}} \right)$ $\frac{c_B}{q_B} = \frac{1}{\xi} \left(c_B + K_{B,A} c_A \right)$	[167]

	Monovalent and bivalent ions exchange model	$\frac{c_A}{q_A} = \frac{1}{\xi_{mono}} \left(c_A + \frac{2}{K_{B,A}} \frac{q_A c_B}{c_A} \right)$ $\frac{c_B}{q_B} = \frac{1}{\xi_{mono}} \left(2c_B + \frac{1}{K_{A,B}} \frac{c_A^2}{(\xi_{mono} - 2q_B)} \right)$	[168]
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